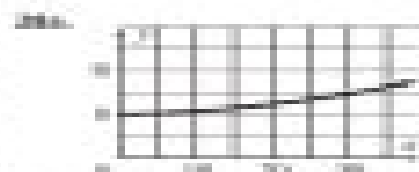
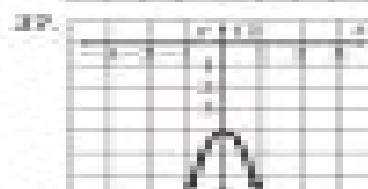
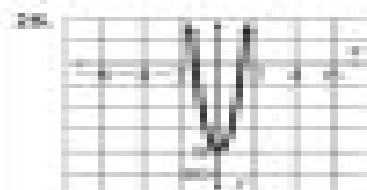
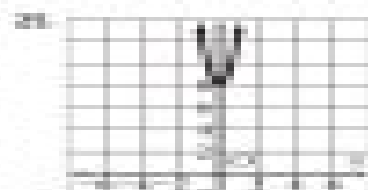
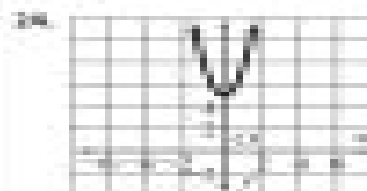
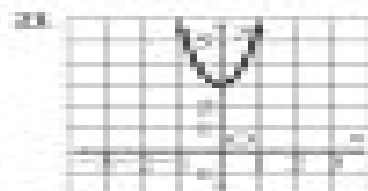
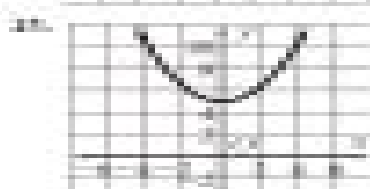
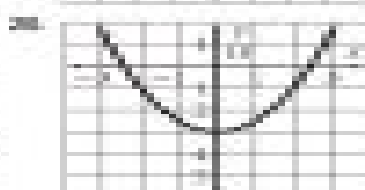
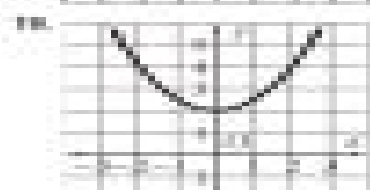
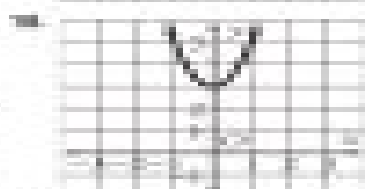
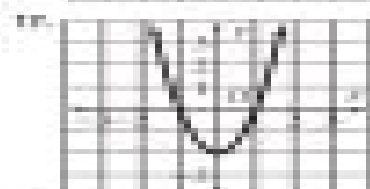
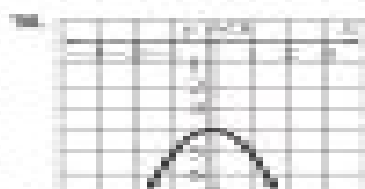
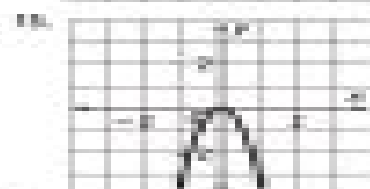
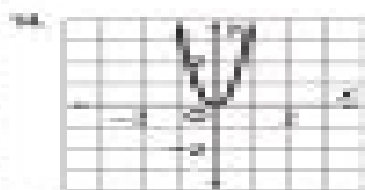
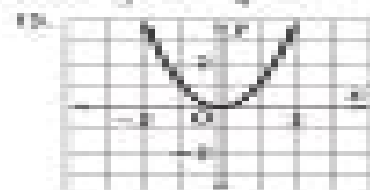


Chapter 10 Answers

Practice 10-1

1. (1,0) minimum 2. (1,0) minimum 3. (1,0) minimum
 4. (0,0) minimum 5. (0,0) minimum 6. (0,0) minimum
 7. $x = 1$, $y = 1$; $x = 3$, $y = 3$ 8. $x = \frac{1}{2}$, $y = \frac{1}{2}$; $x = 1$, $y = 1$
 9. $x = 2$, $y = 2$; $x = 4$, $y = 4$ 10. $x = 1$, $y = 1$; $x = 3$, $y = 3$
 11. $x = 1$, $y = 1$; $x = 3$, $y = 3$ 12. $x = \frac{1}{2}$, $y = \frac{1}{2}$; $x = 1$, $y = 1$



28b. Answers may vary. Sample: Domain should be nonnegative values. Month cannot be negative. 28c. Answers may vary. Sample: Range should be greater than or equal to 0. The minimum value is 0.



29a. Answers may vary. Sample: Domain should be nonnegative values less than or equal to 24. Length of the photograph cannot be negative and cannot be greater than 24. 29b. Answers may vary. Sample: Range should be between 2nd and 1st. There are the minimum and maximum prices available for printing.

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sample answer sugar has a much greater solubility in hot water than cold water so its solubility curve probably resembles the sloping solubility curve of potassium nitrate temperature solubility g 100 ml 10 c 36 g 100 ml 20 c 36 g 100 ml 30 c 37 g 100 ml 40 c 37 g 100 ml 50 c 38 g 100 ml 60 c 38 g 100 ml 70 c 39 g 100 ml 80

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solubility is defined as the maximum quantity of solute that can dissolve in a certain quantity of solvent at a specified temperature differences in the solubility of different solutes are related primarily to differences in the chemical nature of the solutes and the solvents

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1 the solubility of a solute is the maximum mass of the solute that will dissolve in a certain amount of water at a certain this is the same as saying that solubility is the concentration of a unsaturated saturated solution of the solute 2 from your graph what mass of kcl can be dissolved in 100 g of water at these temperatures

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