



# CHEMISTRY

## CHEAT SHEET



# NAMING SIMPLE COMPOUNDS

## NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas  
(Show the steps for naming the compound)



**Steps for Naming:**

1. Identify the cation (metal) and anion (non-metal).
2. Write the name of the cation first, followed by the name of the anion.
3. Use the appropriate prefix for the number of atoms of each element.
4. The cation is named first, followed by the anion.
5. The anion is named second, followed by the cation.
6. The name of the compound is Carbon Dioxide.

### Prefixes

Prefix	Number
mono-	1
di-	2
tri-	3
tetra-	4
penta-	5
hexa-	6
hepta-	7
octa-	8
nona-	9
deca-	10

### Common Names for Simple Compounds

Formula	Name
CO	Carbon Monoxide
CO <sub>2</sub>	Carbon Dioxide
NO	Nitric Oxide
NO <sub>2</sub>	Nitrogen Dioxide
SO <sub>2</sub>	Sulfur Dioxide
SO <sub>3</sub>	Sulfur Trioxide
SiO <sub>2</sub>	Silicon Dioxide
Al <sub>2</sub> O <sub>3</sub>	Aluminum Oxide
Fe <sub>2</sub> O <sub>3</sub>	Iron(III) Oxide
FeO	Iron(II) Oxide

Naming Compounds with a Chemical Formula



Carbon Dioxide

**Steps for Naming:**

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### Prefixes

Prefix	Number	Prefix	Number
mono-	1	deca-	10
di-	2	hepta-	7
tri-	3	octa-	8
tetra-	4	nona-	9
penta-	5	deca-	10
hexa-	6	deca-	10

### Common Names for Simple Compounds

Formula	Name	Formula	Name
CO	Carbon Monoxide	NO <sub>2</sub>	Nitrogen Dioxide
CO <sub>2</sub>	Carbon Dioxide	SO <sub>2</sub>	Sulfur Dioxide
NO	Nitric Oxide	SO <sub>3</sub>	Sulfur Trioxide
NO <sub>2</sub>	Nitrogen Dioxide	SiO <sub>2</sub>	Silicon Dioxide

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**Ying-Ying Zheng**



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