



Refer to the exhibit. The switch and the hub have default configurations, and the switch has built its CDP table. Which of the hosts will capture a copy of the frame when workstation A sends a unicast packet to workstation C?

- ☒ workstation C
- ☐ workstations B and C
- ☐ workstations A, B, C, and the interfaces of the router
- ☐ workstations B, C, D, E, F, and interfaces of the router

8. If a network administrator enters these commands on a switch, what will be the result?

Switch#configure line console 0

Switch#configure line 0 password cisco

Switch#configure line 0 login

- ☐ to secure the console port with the password "cisco"
- ☐ to deny access to the console port by specifying 0 lines are available
- ☐ to gain access to line configuration mode by supplying the required password
- ☒ to configure the privilege exec password that will be used for remote access

10. Which two statements are true about EXEC mode passwords? (Choose two.)

- ☐ The enable secret password command stores the configured password in plain text.
- ☒ The enable secret password command provides better security than the enable password.
- ☒ The enable password and enable secret password protect access to privileged EXEC mode.
- ☐ The service password-encryption command is required to encrypt the enable secret password.
- ☐ Best practices require both the enable password and enable secret password to be configured and used simultaneously.

# Ccna 2 Chapter 3 Answers

**David Minutella, Jeremy Cioara, Heather  
Stevenson**



## Ccna 2 Chapter 3 Answers:

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 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$   
 one mole of HCl would be fully neutralized by one mole of NaOH

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak  $pK_a$  of an unknown acid or  $pK_b$  of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web this point of titration is referred to as the equivalence point considering the 1:1 stoichiometry of this acid base reaction  $\text{NaOH}$  aq  $\text{C}_6\text{H}_8\text{O}_7$  aq  $\text{C}_6\text{H}_7\text{O}_7^-$  aq  $\text{H}_2\text{O}$  l the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the

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