

Matter and Temperature Study Guide

10 pages including KEY

Name: _____ Date: _____

MATTER AND TEMPERATURE STUDY GUIDE

- State the first Law of Thermodynamics.
Energy cannot be created or destroyed, only transformed or changed.
- Define temperature.
Temperature is a measure of the average kinetic energy of the particles in a substance.
- How does temperature relate to kinetic theory?
According to Kinetic Theory, temperature is directly proportional to the kinetic energy of the particles in a substance.
- Convert 25°F to degrees Celsius.







$$^{\circ}\text{C} = \frac{^{\circ}\text{F} - 32.0}{1.8} \rightarrow ^{\circ}\text{C} = \frac{25 - 32.0}{1.8} \rightarrow ^{\circ}\text{C} = \frac{-7.0}{1.8} \rightarrow ^{\circ}\text{C} = -3.9$$

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KEY

- Define Specific Heat.
Specific heat is the amount of heat required to raise the temperature of one gram of a substance by one degree Celsius.
- What is conduction?
Conduction is the transfer of heat through direct contact between materials.
- What is convection?
Convection is the transfer of heat by the movement of fluids (liquids or gases).
- What is radiation?
Radiation is the transfer of energy through electromagnetic waves.
- How are conduction, convection, and radiation similar?
They are all methods of heat transfer.

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- How do convection currents work?
Convection currents occur when warmer, less dense fluid rises and cooler, denser fluid sinks, creating a cycle.
- What is a conductor?
A material that allows heat or electricity to flow through it easily.
- What is an insulator?
A material that resists the flow of heat or electricity.
- What type of material makes the best conductor?
Metals, such as copper and aluminum.
- What type of materials make the best insulators?
Materials like rubber, glass, and wood.
- What is the volume of a substance?
165 g?
- What is the density of a substance?


- What is the mass of a substance?


- What is the volume of a gas?



Name: _____ Date: _____

- What is a chemical change?
A change that results in the formation of new chemical substances.
- Give 3 examples of chemical changes.
Answers will vary. Some examples: rusting of iron, burning of wood, and baking a cake.
- What is a physical change?
A change in which no new substances are formed and the change is often reversible.
- Give 3 examples of physical changes.
Answers will vary. Some examples: melting of ice, boiling of water, and dissolving sugar.

Name: _____ Date: _____

KEY

- State the Law of Conservation of Matter.
Matter is neither created nor destroyed in a chemical reaction.
- When liquid water is put into the freezer, what happens to it?
It freezes and becomes solid ice.
- If you throw pennies into a fountain and they sink to the bottom, how does the density of the penny compare to the density of the water in the fountain?
The density of the penny is greater than the density of the water.
- What state of matter has no definite shape and is made up of charged particles?
Plasma.
- If the volume of a gas increases, what happens to the temperature?
The temperature decreases (assuming pressure is constant).
- Question #42 is an example of what state of matter?
Gas.

Composition Structure And Temperature Study Guide Answers

IM Harris



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