

SECTION - B

Note : There are 12 Short Answer Type Questions from Q. 3 to Q. 14 carrying two marks each. You have to attempt any eight questions. This section includes ten theory questions and two numerical problems. Write appropriate units for each calculated value.

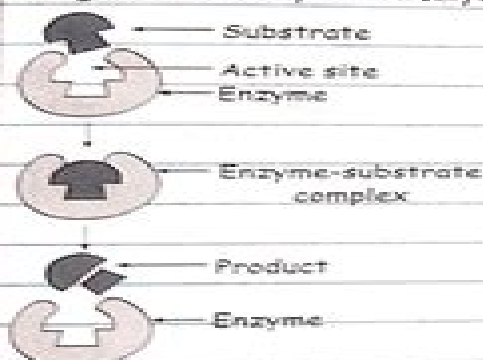
- Q. 3. (a) It oxidises lead sulphide (PbS) to lead sulphate (PbSO_4) changing oxidation state of S from -2 to $+6$.



- (b) Ozone oxidises nitrogen oxide (nitric oxide) to nitrogen dioxide.



- Q. 4. Diagram of enzyme catalysis :



Enzyme catalysis

(2 mark)

- Q. 5. If x_1 and x_2 are the mole fractions of solvent and solute respectively, then

$$x_1 + x_2 = 1$$

By Raoult's law,

$$P = x_1 \times P_0$$

where P^0 is the vapour pressure of a pure solvent and P is the vapour pressure of the solution at given temperature.

$$\therefore \frac{P}{P_0} = x_1$$

$$1 - \frac{P}{P_0} = 1 - x_1 \quad \therefore \frac{P_0 - P}{P_0} = x_2$$

$$\text{If } P_0 - P = \Delta P, \text{ then } \frac{\Delta P}{P_0} = x_2.$$

($\frac{1}{2}$ mark)

($\frac{1}{2}$ mark)

(1 mark)

Cybered Chemistry Answer Guide

Leonard W. Fine



Cybered Chemistry Answer Guide:

Absolute Beginner's Guide to Computer Basics Michael Miller, 2004 From CD burners to digital cameras and PDAs this non technical illustrated book shows new computers users how to set up their computer hardware find their way around the Windows XP desktop learn to use essential software programs and get connected to and start using the Internet

Illinois Chemistry Teacher ,2001 *The Science Teacher* ,2000 Library Media Connection ,1999 **The American Biology Teacher** ,2000 Study Guide and Selected Solutions Manual for Basic Chemistry Karen C. Timberlake, William Timberlake, 2019-06-06 The Study Guide and Selected Solutions Manual as written specifically to assist students using Basic Chemistry It contains learning objectives chapter outlines additional problems with self tests and answers and answers to the odd numbered problems in the text **Solutions Guide for Chemistry** Steven S. Zumdahl, Kenneth C. Brooks, 1989 **The Chemistry Problem Solver** ,2001 Study Guide and Selected Solutions Manual for Chemistry for Changing Times John W. Hill, Richard Jones, Doris K. Kolb, Terry W. McCreary, Marcia L. Gillette, 2012-01-31 *Chemistry Reference Guide (RES)* Responsive Education Solutions, 2013-08-01 Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions **Chemistry** National Learning Corporation, 2018 The Chemistry Passbook R prepares you for your test by allowing you to take practice exams in the subjects you need to study It provides hundreds of questions and answers in the areas that will likely be covered on your upcoming exam Chemistry Equations & Answers (Speedy Study Guide) Speedy Publishing LLC, 2014-11-19 Chemistry can be one of the more difficult subjects in school mainly because memorization is crucial It can be hard to memorize tons of facts and equations A chemistry equations and answers guide does help students some more than others You can refer to the study guide to determine which formula is suitable to use for the question at hand The more savvy student might consider one or more ways to creatively benefit from the guide For example creating useful tricks to aid you during a challenging test or quiz Chemistry may seem difficult but there are useful study methods **Student's Study Guide and Selected Solution Manual for Chemistry for Changing Times** John W. Hill, Terry W. McCreary, Doris K. Kolb, Richard Jones, Marcia Gillette, 2015-04-02 The Study Guide and Selected Solutions Manual assists students with the text material It contains learning objectives chapter outlines additional problems with self tests and answers and answers to the odd numbered problems in the text *Answer Guide for Chemistry, 2nd Edition* Leonard W. Fine, 1978 *Chemistry Equations & Answers* Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to

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Question and Answer Guide to Introductory Chemistry Jerome R. Hofeld, 1952 **Chemistry Revised Answer Guide** Steven S. Zumdahl, **Student Solutions Manual for General Chemistry** John McMurry, Robert Fay, 2013-03-07

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