

# 1

## D-Link Antenna-kit Installation Guide

### Step 1

Remove the default antenna.



### Step 2

Connect the extension cable onto the external connector on device.



### Step 3



Install the surge protector as recommended position when you do not purchase the low loss cables (ANT24-CB03N, ANT24-CB06N and ANT24-CB09N...)

### Step 4



Install the surge protector as recommended position near the device side, when you have the low loss cables (ANT24-CB03N, ANT24-CB06N and ANT24-CB09N ...)

### Step 5

Use water-proof tape to wrap around both connectors tightly, as the pictures showed below



When install the surge protector between cables, wrap the connectors as well. The lightning protection adaptor for grounding the antenna.



With Surge protector (ANT24-SP) adaptor on an outdoor serial is attached to the lightning protection system (ground) of the building. All devices downstream are then protected by the ground from lightning strikes. It can protect your sensitive WLAN equipments from high voltage surges caused by discharge and transients at the antennas. It will be integrated in the cabling system between the antenna and the coax-cable and must have direct ground contact.

# D Link Ant24 18manual

**L Manion**



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web solution 1 the two half reactions balanced as if in acidic solution  $2\text{CN}^{-} + \text{Au} + \text{CN}^{-} + 2\text{e}^{-} + 2\text{H}^{+} + \text{O}_2 \rightarrow \text{H}_2\text{O}_2 + 2\text{HCN}$  make electrons equal convert to basic solution  $4\text{CN}^{-} + 2\text{Au} + 2\text{OH}^{-} + 2\text{e}^{-} \rightarrow 2\text{AuCN}^{-} + 2\text{H}_2\text{O} + \text{O}_2 + 2\text{OH}^{-}$  the final answer

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web balance each half reaction in basic solution 4 5 6 7  $\text{Cr}_2\text{O}_7^{2-}$   $\text{NO}_3^-$   $\text{SO}_4^{2-}$   $\text{MnO}_2$   $\text{Cr}_3^{+}$   $\text{SO}_2$   $\text{Mn}_2\text{O}_3$  balance each redox reaction in acid solution using the half reaction method 8  $\text{H}_2\text{O}_2$   $\text{Cr}_2\text{O}_7^{2-}$  2 9  $\text{TeO}_3$  2  $\text{N}_2\text{O}_4$   $\text{O}_2$   $\text{Cr}_3^{+}$  10  $\text{ReO}_4^-$  11  $\text{PbO}_2$   $\text{IO}_3^-$   $\text{H}_2\text{AsO}_4^-$  12  $\text{AsO}_3^{3-}$   $\text{IO}_3^-$   $\text{H}_2\text{AsO}_4^-$

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively  $\text{C}_6\text{H}_8\text{O}_6$   $2\text{H}_2\text{NO}_2$   $\text{C}_6\text{H}_6\text{O}_6$   $2\text{H}_2\text{O}_2$   $\text{NO}$  choose 1 answer  $\text{C}_6\text{H}_8\text{O}_6$  and  $\text{NO}_2$  a  $\text{C}_6\text{H}_8\text{O}_6$  and  $\text{NO}_2$  b  $\text{C}_6\text{H}_8\text{O}_6$  and  $\text{NO}_2$  c  $\text{C}_6\text{H}_8\text{O}_6$  and  $\text{NO}_2$  d  $\text{C}_6\text{H}_8\text{O}_6$  and  $\text{NO}_2$

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