#### Chapter 16 (Pi Systems) Homework, CHEM 224 NAME \_

Draw the structures of the following compounds

a) cyclopentadiene (explain why the numbers for the double bonds (1,3-), which are formally required, are usually omitted; i.e., why is only one isomer possible; why not 1,2- or 1,4-cyclopentadiene?

b) 1,3,5-cycloheptatriene

c) (2E, 4Z)-2,4-heptadiene

d) isoprene (look it up and also give its IUPAC name)

Draw a molecule that has the following characteristics.

 a) a decalin (2 six-member rings joined side by side) with both conjugated and non-conjugated double bond (three double bond total)

b) a compound containing a 5-member ring and conjugated doubles bond that are locked in an s-trans orientation

c) an eight-member ring containing two cumulated double bonds (i.e., an allene)

d) a decalin with two double bonds that are locked in an s-cis orientation

#### 3.

a) Draw isoprene in both the s-cis and the s-trans confirmation

b) Draw all trans and all cis 2,4,6-octatriene

 c) Draw all trans 1,3,5,7-octatetraene in all possible confirmations (rotamers, not geometric isomers) where the four double bonds are all conjugated (planar)

#### 4.

a) Fill in the missing diene that would lead to this isomer as the thermodynamic, but not the kinetic, product.

b) Show the mechanism for the above reaction, including the formation of both kinetic (low temperature) and thermodynamic (higher temperature) products.

c) Show the mechanism for the conversion of the kinetic to the thermodynamic product.

d) Show a reasonable energy curve for this transformation and draw in all structures, including intermediates and transition states

5. Draw an example of the following.

a) an acyclic, conjugated diene that cannot be s-cis due to substitution at C2 and C3 of the diene

b) an acyclic, conjugated diene that cannot be s-cis due to substitution at C1 and C4 of the diene

Rank the following alkenes from least reactive with 1,3-butadiene to most reactive. Explain the ranking of each compound with a couple of words.



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**Jack Hipple** 

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