

Class 4.2

SECTION 3

Write an equation for each distribution operation in the space provided.

1. _____ The multiplication of a gas on RST is distributive of the gas.
 (a) multiplied by the constant 1 mol. (b) multiplied by 12.0 L.
 (c) divided by the constant 1 mol. (d) divided by 12.0 L.

2. _____ (a) Insert appropriate Δ or $\frac{\Delta}{\Delta}$ subscript that

following will cover the volume in liters:

- (a) increasing Δ or $\frac{\Delta}{\Delta}$ increasing Δ
 (b) decreasing Δ or $\frac{\Delta}{\Delta}$ decreasing Δ

3. _____ From equal flasks, A and B, measure different gases of equal volume in the same temperature conditions.

- (a) The two flasks show different air equal number of molecules.
 True or False? True

- (b) The two cylinders contain equal masses. True or False?
 False

Write each of the correct answer in the box. Show all calculations in the space provided.

4. _____ List the chemical formulae in order of the following properties.

Formula	Molar mass (g/mol)
<u> </u>	<u> </u>
<u> </u>	<u> </u>
<u> </u>	<u> </u>
<u> </u>	<u> </u>
<u> </u>	<u> </u>

5. Rank the gases in order of

- (a) Diffusion rate (assuming the same conditions as 4.2) (smallest to largest)
 the gases
 (b) Diffusion rate (assuming 1 bar)
 (c) Diffusion rate (assuming the same density) 1.0 g/L
 (d) Diffusion rate (at 1% concentration) 100% 1.0 g/L

Chapter 11 Review Gases Section 3 Short Answer

Padhraic Smyth



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Cracking the AP Physics B Exam, 2014 Edition* Steven A. Leduc, John J. Miller, 2013-09-03 Presents a study plan to build knowledge and confidence discusses study skills and strategies reviews core topics and provides two full length practice tests **Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity Holt Chemistry Ralph Thomas Myers, 2004 The Pearson Complete Guide For Aiee 2/e Khattar, **Telegraphic Journal and Monthly Illustrated Review of Electrical Science**, 1925 Earth's Surface: Teacher's ed, 2005 **The Electrical Review**, 1925 **The Electrician**, 1908 **Chemistry** Henry Dorin, 1987 *The Spectator*, 1876 **The Chemical News and Journal of Industrial Science**, 1913 **Bibliography of Bibliographies on Chemistry and Chemical Technology, 1900-1924**, 1925 **The Minimum Protein Requirements of Cattle** National Research Council (U.S.). Committee on Animal Nutrition, Harold Hanson Mitchell, 1929 *Bulletin of the National Research Council* National Research Council (U.S.). Research Information Service, 1929 **Bulletin** National Research Council (U.S.), 1929 **Bibliography of Bibliographies on Chemistry and Chemical Technology, 1900-1924** National Research Council (U.S.). Research Information Service, 1929 *Addison-Wesley Introduction to Physical Science* Michael B. Leyden, 1988 "Code of Massachusetts regulations, 1994", 1994 Archival snapshot of entire looseleaf Code of Massachusetts Regulations held by the Social Law Library of Massachusetts as of January 2020 The New York Times Index, 1928

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