

Concept Review

Section 8-1: Covalent Bonds

Answer the following items in the space provided.

1. What does a molecular bond-forming reaction look like?
2. Why is the H_2 molecule more stable than two separate hydrogen atoms?
3. Explain why the stability difference in H_2 does not show that bond-forming is more energetically favorable.
4. What does a molecular bond-forming reaction look like?
5. What is enthalpy change?
6. Describe potential energy changes that occur when two hydrogen atoms approach each other, what forms a covalent bond?
7. In terms of energy, why is the H_2 bond stable?
8. Why is bond length an average value? How is bond energy?
9. Describe the physical properties of molecules that the covalent bond, ionic, and covalent bonds.

Complete each statement below by choosing a term from the following list. Name may be used more than once.

- | | | | |
|---------------------|---------------------|----------|----------------|
| strategy | electron-attraction | electron | ionic covalent |
| electron-attraction | electron | electron | electron |
1. Bond _____ is the force that holds atoms together. Bonding is the force that holds molecules together.
 2. Bond _____ is the force that holds atoms together. Bonding is the force that holds molecules together.
 3. The attraction of an atom for electrons is called the _____.
 4. In _____, electrons are attracted to the nucleus of the atom. In ionic bonding, electrons are attracted to the nucleus of the atom.
 5. In _____, electrons are attracted to the nucleus of the atom. In covalent bonding, electrons are attracted to the nucleus of the atom.

Concept Review Chapter 6 Section Covalent Bonds

Answer Key

N Noddings



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Holt Chemistry Ralph Thomas Myers, 2004 Organic Chemistry Michael B. Smith, 2010-10-18 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules **Principles of Cell Biology** George Plopper, 2014-10-21 Written for undergraduate cell biology courses Principles of Cell Biology Second Edition provides students with the formula for understanding the fundamental concepts of cell biology This practical text focuses on the underlying principles that illustrate both how cells function as well as how we study them It identifies 10 specific principles of cell biology and devotes a separate chapter to illustrate each The result is a shift away from the traditional focus on technical details and towards a more integrative view of cellular activity that is flexible and can be tailored to suit students with a broad range of backgrounds CHEMISTRY SILBERBERG, 2003 **Human Biology** Daniel Chiras, 2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course **Class 8-12 Chemistry Questions and Answers PDF** Arshad Iqbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class questions Class 8

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