Chemistry Study Guide



Objective:

To understand the structure of an atom, and the nature + properties of subatomic particles 🛫

Matter:

Anything that has mass and occupies space 💥 🕥

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the smallest, indivisible particle &.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles

Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|----------|--------|------------|--------|-------------|
| Proton | p." | 1 | -1 | Nucleus |
| Neutron | m | 1 | 0 | Nucleus |
| Electron | • | -0 | -1 | Orbit/Shell |

Conceptual Chemistry Study Guide

Herbert Levin

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