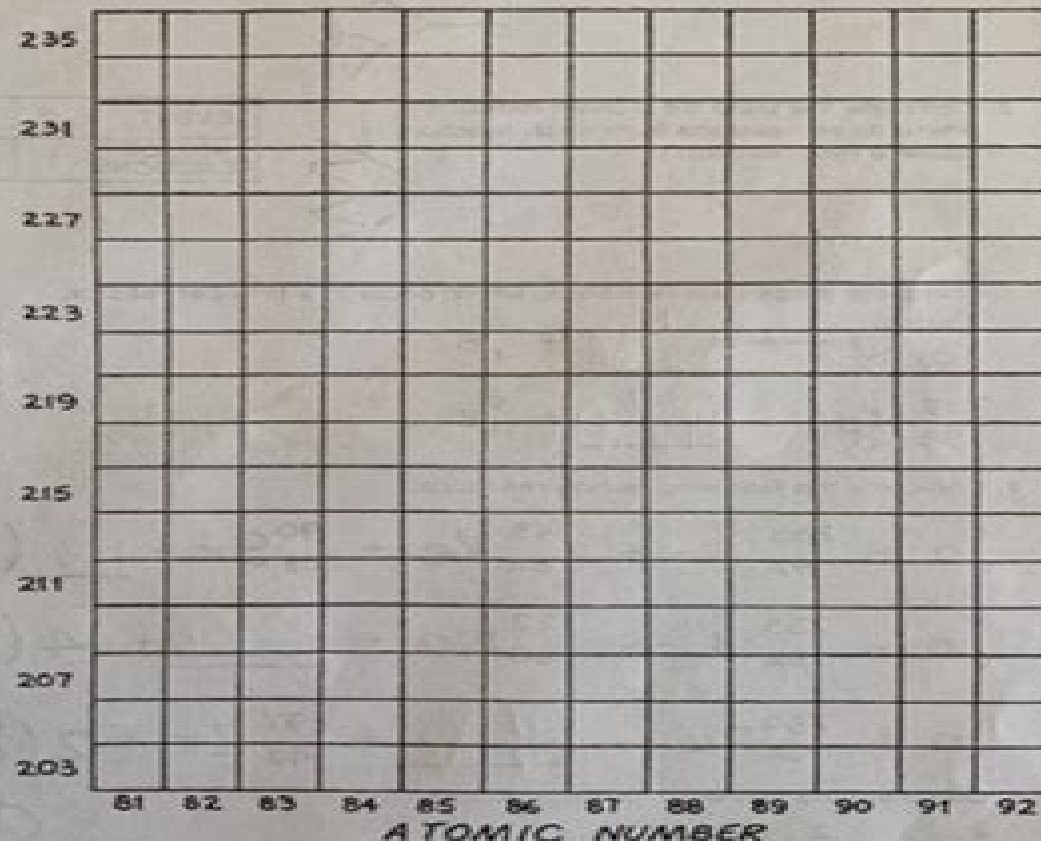


**CONCEPTUAL Physics** PRACTICE PAGE**Chapter 33 The Atomic Nucleus and Radioactivity**  
**Natural Transmutation**

Fill in the decay-scheme diagram below, similar to that shown in Figure 33.22 in your textbook, but beginning with U-235 and ending with an isotope of lead. Use the table at the left and identify each element in the series with its chemical symbol.

Step	Particle emitted
1	Alpha
2	Beta
3	Alpha
4	Alpha
5	Beta
6	Alpha
7	Alpha
8	Alpha
9	Beta
10	Alpha
11	Beta
12	Stable

MASS NUMBER



What isotope is the final product?

\_\_\_\_\_

# Conceptual Physics Chapter 33 Answers

**Paul G. Hewitt**



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