### Bendina Basics

#### ANSWER KEY

#### Section A: Complete the chart using a periodic table to help you.

Element	Atomic Symbol	Total # of Electrons	# of Valence Electrons	# of Electrons Needed to Gain or Lose to Fill Outer Shells	Oxidation Number
Chlorine	Cl	17	7	Gain 1	1-
Potassium	к	19	1	Lose I	1+
Magnesium	Mg	12	2	Lose 2	2+
Fluorine	F	9	7	Gain I	1-
Aluminum	Al	13	3	Lose 3	3+
Sodium	Na	11	1	Lose 1	1+
Nitrogen	N	14	4	Gain 3	3-
Oxygen	0	8	6	Gain 2	2-
Hydrogen	н	1	1	Gain or Lose 1	1+ or 1-
Carbon	C	6	4:	Gain or Lose 4	4+ or 4-
Lodine	1	53	7	Gain 1	1-

#### Answer these questions:

- An atom that gains one or more electrons will have a NEGATIVE charge.
- An atom that loses one or more electrons will have a POSTIVE charge.
- An atom that gains or loses one or more electrons is called an ION.
- A positive ion is called a CATION and a negative ion is called an ANION.

# **Bonding Basics Review Answer Key**

**Arshad Iqbal** 

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