Name:	Date:	Class period

Unit Test: Chemical Bonding (A) ANSWER KEY

100

Part I: Matching (2 points each) Please match the term with the correct definition. There will be extra choices and some choices may be used more than once.

A	Electronegativity	F. Molecular Formula	K. Crystal
	Covalent bonding	G. Linear	L. Pyramid
	Ductile	H. Non-polar	M. Polar
D.	Agueous	I. Solid	N. VSEPR
E.	Tetrahedral	 J. Trigonal bipyramidal 	O. Malleable
	Octet Rule	Q. Ionic bonding	R. Empirical Formula

- 1. _K_ A 3-D, repeating pattern of + and ions, formed by ionic compound
- _A_ Tendency for an atom to attract electrons when chemically bonded to another atom
- B A sharing of electrons
- _P__ Atoms will gain or lose enough electrons in order to become isoelectronic with a noble gas
- Q A transfer of electrons
- 6. __R__ A chemical formula that is arranged in the smallest whole number ratio
- 7. __D_ The term that means dissolved in water
- 8. __F__ A chemical formula that describes the makeup of a single molecule
- The shape (geometry) that is always an exception to the octet rule
- 10. __M_ A bond where electrons are shared unequally between atoms
- 11. L One of the shapes (geometries) that will always be polar
- 12. __H__ A bond where electrons are shared equally between atoms

Part II: Multiple Choice (2 points each) Please circle the <u>most correct</u> answer. Write your answer on the given lines.

- 1. E A triple covalent bond shares electrons between atoms.
 a. two
 b. three
 c. four
 d. five
 e. six
 f. Unable to tell

Bonding Test Answer Key

V A Coveney

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