

Concept Review

Section: Compound Names and Formulas

1. **Explain** the difference between iron(II) nitrate and iron(III) nitrate. What is the significance of the Roman numerals?

2. **Name** the following ionic compounds, keeping in mind that a transition metal cation must include its charge.

- _____ a. TiO_2
- _____ b. BaCl_2
- _____ c. CuCl_3
- _____ d. KI
- _____ e. SrCl_2
- _____ f. CuBr_2

3. **Describe** how covalent compounds are named.

4. **Write** the chemical formulas for the following compounds:

- _____ a. lithium oxide (ionic)
- _____ b. carbon monoxide (covalent)
- _____ c. carbon tetrachloride (covalent)
- _____ d. nitrogen trifluoride (covalent)
- _____ e. calcium chloride (ionic)

5. **Contrast** molecular formulas and empirical formulas.

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M Lipman



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