

COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

electrons in the middle share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

wanted available share

| | | |
|---|--|---|
| <p>1. $\text{H} + \text{H} (\text{H}_2)$</p> <p>W - 4</p> <p>A - 2</p> <p>S - 2</p> | <p><i>wanted 2 + 2</i></p> <p><i>available</i></p> <p>$\text{H} \cdot \cdot \text{H}$</p> <p>$\text{H} \cdot \quad \cdot \text{H}$</p> | <p>$\text{H} - \text{H}$</p> <p><i>single bond = share 2 e⁻</i></p> |
| <p>2. $\text{F} + \text{F} (\text{F}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 7 + 7 = 14</p> <p>S - 2</p> | <p>$\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot$</p> <p><i>2 e⁻</i></p> | <p>$\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot$</p> <p><i>single bond 2 e⁻</i></p> |
| <p>3. $\text{O} + \text{O} (\text{O}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 6 + 6 = 12</p> <p>S - 4</p> | <p>$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot$</p> <p><i>4 e⁻</i></p> | <p>$\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot$</p> <p><i>4 double bond</i></p> |
| <p>4. $\text{N} + \text{N} (\text{N}_2)$</p> <p>W - 8 + 8 = 16</p> <p>A - 5 + 5 = 10</p> <p>S - 6</p> | <p>$\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot$</p> | <p>$\cdot \ddot{\text{N}} \equiv \ddot{\text{N}} \cdot$</p> |
| <p>5. $\text{C} + \text{O} (\text{CO}_2)$</p> <p>W - 8 + 8 + 8 = 24</p> <p>A - 4 + 6 + 6 = 16</p> <p>S - 8</p> | <p>$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot$</p> | <p>$\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot$</p> |
| <p>6. $\text{H} + \text{O} (\text{H}_2\text{O})$</p> <p>W - 2 + 2 + 8 = 12</p> <p>A - 1 + 1 + 6 = 8</p> <p>S - 4</p> | <p>$\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}$</p> | <p>$\text{H} - \ddot{\text{O}} - \text{H}$</p> |

Chap09 Answers Covalent Bonding

SJ Ball



Chap09 Answers Covalent Bonding:

This book delves into Chap09 Answers Covalent Bonding. Chap09 Answers Covalent Bonding is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chap09 Answers Covalent Bonding, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 4: Chap09 Answers Covalent Bonding in Specific Contexts
 - Chapter 5: Conclusion
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