

CHEMICAL QUANTITIES

Practice Problems

On a piece of paper, solve the following problems.

YOU MUST SHOW YOUR WORK TO GET CREDIT FOR YOUR ANSWERS!!!

SECTION 10.1 THE MOLE: A MEASUREMENT OF MATTER

- What is the molar mass of sucrose (C₁₂H₂₂O₁₁)?
- 2. What is the molar mass of each of the following compounds?
 - a, phosphorus pentachloride (PCI-)
 - b. uranium hexafluoride (UF₄)
- 3. Calculate the molar mass of each of the following ionic compounds:
 - a. KMinO₄
 - b. Cas(PO₄):
- 4. How many moles is 3.52 × 1024 molecules of water?
- 5. How many atoms of zinc are in 0.60 mol of zinc?
- 6. What is the mass of 1.00 mol of exygen (O2)?

SECTION 10.2 MOLE-MASS AND MOLE-VOLUME RELATIONSHIPS

- 1. What is the molar mass of each of the following compounds?
 - a. CaHagOa

b. NaHCOs.

e. Calling

d. KNH4SO₄

- 2. Calculate the mass in grams of each of the following:
 - a. 8.0 mol lead oxide (PbO).

- d. 1.50 × 10⁻² mol molecular oxygen (O₅)
- b. 0.75 mol hydrogen sulfide (H₂S)

- e. 2.30 mol ethylene glycol (C₂H₂O₂)
- e. 0.00100 mol silicon tetrahydride (SiH₄).
- How many grams are in 1.73 mol of dinitrogen pentoxide (N₂O₂)?
- How many grams are in 0.658 mol of calcium phosphate [Ca₂(PO₄)₂]?
- 5. Calculate the number of moles in each of the following:
 - a. 0.50 g sodium bromide (NaBr).

- d, 0.00100 g monochloromethane (CH₂CI)
- b. 13.5 g magnesium nitrate [Mg(NO₁)₂]
- e. 1.50 × 10⁻³ g propylene glycol [C₃H₄(OH)₂]
- e. 1.02 g magnesium chloride (MgCl₂)
- 6. A chemist plans to use 435.0 grams of ammonium nitrate (NH₄NO₅) in a reaction. How many moles of the compound is this?
- 7. A solution is to be prepared in a laboratory. The solution requires 0.0465 mol of quinine (C₂₀H₂₄N₂O₂). What mass, in grams, should the laboratory technician obtain in order to make the solution?
- 8. What is the volume at STP of 2.66 mol of methane (CHa) gas?
- 9. How many moles is 135 L of ammonia (NH₃) gas at STP?

Chapter 10 Chemical Quantities Study Guide Answer Key

Mr. Rohit Manglik

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web sep 21 2015 synopsis when jane doe is found naked in times square with her memory erased and her body covered in fresh tattoos agent kurt weller and his fbi team are instantly drawn to the case together

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