

Section 6-1

Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything else is missing some

Concept Review Section Simple Ions Answers

RC Schank



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Wooden,Randy Blazak,2001 Topics covered include exploring boundary between deviance and criminality in the lives of young people who are deeply involved in the youth culture show how youth culture is not a set of categories so much as it is a dynamic and creative response to the confusions of growing up in modern society The Classical Review ,1924

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Contemporary Chemical Analysis Judith F. Rubinson,Kenneth A. Rubinson,1998 This book provides a readable yet rigorous introduction to analytical methods with a focus on problem solving skills It stresses the fundamental concepts of chemical analysis and through examples from current journals and other science media shows how the principles and practice of analytical chemistry are used to produce answers to questions in all areas of scientific study and practice Features a balance of topics that is closer to contemporary analytical practice than those covered by other books Introduces the tools that are ubiquitous in analytical chemistry e g statistics sampling and sample preparation Discusses methods depending on chemical kinetics which are so widely used in medicine and biology Features a number of problems that call for the use of a spreadsheet to generate data which is then plotted to show trends Includes answers for all numerical problems in an appendix Mathematical Reviews ,2007

Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *Ion Exchange and Solvent Extraction* Yitzhak Marcus,Arup K. SenGupta,2004-07-23 Volume 17 in the Ion Exchange and Solvent Extraction series represents the vanguard of research on solvent extraction It covers the principles of electrolyte extraction and other subjects of increasing interest to the field This

volume begins with pharmaceutical applications of supercritical fluid solvents particularly supercritical carbon dioxide It also contains chapters on liquid ion exchangers and relevant experiment protocols SCF applications in drug formulation and pollution reduction exploiting SCF as reaction media applications of metal bis dicarbollide in analytical chemistry and radioactive waste treatment and synergistic extraction of metal ions Volume 17 discusses the ion exchange isothermal supersaturation technique metal separation via pH induced parametric pumping modeling of ion exchange kinetics for ultrapure water and the engineering of activated carbons and carbonaceous materials for removal of metal ions and organic micropollutants in water Volume 17 cover topics that include supercritical fluid applications applications of metal bis dicarbollide and synergistic extraction of metal ions *CRC Critical Reviews in Microbiology* ,1972 *Chemical Reviews* ,1928 Features review articles covering key areas of research and progress This journal provides comprehensive and expert critical analysis in organic inorganic physical analytical theoretical and biological chemistry *Science Progress* ,1920

Thermodynamics and Kinetics of Water-rock Interaction Eric H. Oelkers, Jacques Schott, 2009 Volume 70 of *Reviews in Mineralogy and Geochemistry* represents an extensive review of the material presented by the invited speakers at a short course on Thermodynamics and Kinetics of Water Rock Interaction held prior to the 19th annual V M Goldschmidt Conference in Davos Switzerland June 19 21 2009 Contents Thermodynamic Databases for Water Rock Interaction Thermodynamics of Solid Solution Aqueous Solution Systems Mineral Replacement Reactions Thermodynamic Concepts in Modeling Sorption at the Mineral Water Interface Surface Complexation Modeling Mineral Fluid Equilibria at the Molecular Scale The Link Between Mineral Dissolution Precipitation Kinetics and Solution Chemistry Organics in Water Rock Interactions Mineral Precipitation Kinetics Towards an Integrated Model of Weathering Climate and Biospheric Processes Approaches to Modeling Weathered Regolith Fluid Rock Interaction A Reactive Transport Approach Geochemical Modeling of Reaction Paths and Geochemical Reaction Networks

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