

Name _____

Class _____

Date _____

Skills Worksheet

Concept Review**Section: Concentration and Molarity**

Complete each statement below by choosing a term from the following list.
Use each term only once.

concentration

moles

solute

solution

liter

parts per million

molality

molarity

ratios

1. _____ is the quantity of solute in a specific quantity of solvent or solution. One way of expressing concentration is to use the unit of measure called _____; it is the ratio of _____ amount (in moles) to _____ volume (in liters), or simply _____ of solute per L of solution. Table 1 in your textbook lists other concentration units or _____, such as _____ ($m = \text{mol solute/kg solvent}$) and _____ ($\text{ppm} = \text{mL solute/1 000 000.0 mL solution}$).

Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?
3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$)?

Concept Review Section Solubility And Concentration

LM Reder



Concept Review Section Solubility And Concentration:

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

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Basic Concepts of Chemistry Malone, 2000-04-18 **Holt Chemistry** Ralph Thomas Myers, 2004 **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 *Aeronautical Engineering Review* ,1957 **Chemical Interactions** ,2005

Organic Chemistry David R. Klein, 2016-12-05 In Organic Chemistry 3rd Edition Dr David Klein builds on the phenomenal success of the first two editions which presented his unique skills based approach to learning organic chemistry Dr Klein s skills based approach includes all of the concepts typically covered in an organic chemistry textbook and places special emphasis on skills development to support these concepts This emphasis on skills development in unique SkillBuilder examples provides extensive opportunities for two semester Organic Chemistry students to develop proficiency in the key skills necessary to succeed in organic chemistry *Proceedings of the Section of Sciences* ,1927 Review of American Chemical Research William Albert Noyes, 1900 *Review of American Chemical Research* ,1897 Chemistry and Industry Review ,1929 **Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson** Norman E. Griswold, Henry Fuller Holtzclaw, 1988 *Robinson Chemistry Study Guide* Robinson, 1992

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