

25.1 Radioactivity

2. α particles ✓

1

(b) (i)

type of radiation	Typical range in air / m
α	0.04 ✓
β	0.40 ✓

Allow students to use their own distance units in the table

α allow 0.03 $\rightarrow$ 0.07 m

β allow 0.20 $\rightarrow$ 3.0 m.

If a range is given in the table use the larger value.

A specific number is required e.g. not just a few cm.

2

- (ii) reference to the inverse square law of (γ radiation)
 or
 reference to lowering of the solid angle (subtended by the detector as it moves away)
 or
 radiation is spread out (over a larger surface area as the detector is moved away) ✓

(owntie)

Ignore any references to other types of radiation.

Any contradiction loses the mark. For example,

follows inverse square law so intensity falls

exponentially.

1

- (c) dust may be ingested / taken into the body / breathed in ✓

First mark for ingestion not just on the body

causing (molecules in human tissue / cells) to be made cancerous / killed / damaged by ionisation ✓

Second mark for idea of damage from ionisation

2

[6]

25.2 Nuclear Decay Equations

1. $a = 1$

Concept Reviews Radioactivity And Nuclear Reactions Answers

RJ Shavelson



Concept Reviews Radioactivity And Nuclear Reactions Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16 Ebook Chemistry: The Molecular Nature of Matter and Change **Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living, relevant science; to encourage learning and critical thinking; and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of general chemistry, as a prep book it promotes active involvement with the material. There are special features throughout that reinforce concepts and help to develop strong problem-solving and study skills. Updated to include an Interactive LearningWare problems CD containing several of the chapter-ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems. *From Atoms to Galaxies* Sadri Hassani, 2011-06-13 College students in the United States are becoming increasingly incapable of differentiating between proven facts delivered by scientific inquiry and the speculations of pseudoscience. In an effort to help stem this disturbing trend, *From Atoms to Galaxies: A Conceptual Physics Approach to Scientific Awareness* teaches heightened scientific acuity as it educates students about the physical world and gives them answers to questions large and small. Written by Sadri Hassani, the author of several mathematical physics textbooks, this work covers the essentials of modern physics in a way that is as thorough as it is compelling and accessible. Some of you might want to know: How did Galileo come to think about the first law of motion? Did Newton actually discover gravity by way of an apple and an accident? Or maybe you have mulled over: Is it possible for Santa Claus to deliver all his toys? Is it possible to prove that Elvis does not visit Graceland every midnight? Or perhaps you've even wondered: If ancient Taoism really parallels modern physics? If psychoanalysis can actually be called a science? How is it that some philosophies of science may imply that a 650-year-old woman can give birth to a child? No Advanced Mathematics Required A primary textbook for undergraduate students not

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