

Section 6-1

Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything
else is missing some

Concept 5 Review Section Simple Ions Answers

Jin-Ying Zhang



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The Handbook of Lithium-Ion Battery Pack Design John T. Warner, 2024-05-14 The Handbook of Lithium Ion Battery Pack Design Chemistry Components Types and Terminology Second Edition provides a clear and concise explanation of EV and Li ion batteries for readers that are new to the field The second edition expands and updates all topics covered in the original book adding more details to all existing chapters and including major updates to align with all of the rapid changes the industry has experienced over the past few years This handbook offers a layman s explanation of the history of vehicle electrification and battery technology describing the various terminology and acronyms and explaining how to do simple calculations that can be used in determining basic battery sizing capacity voltage and energy By the end of this book the reader will have a solid understanding of the terminology around Li ion batteries and be able to undertake simple battery calculations The book is immensely useful to beginning and experienced engineers alike who are moving into the battery field Li ion batteries are one of the most unique systems in automobiles today in that they combine multiple engineering disciplines yet most engineering programs focus on only a single engineering field This book provides the reader with a reference to the history terminology and design criteria needed to understand the Li ion battery and to successfully lay out a new battery concept Whether you are an electrical engineer a mechanical engineer or a chemist this book will help you better appreciate the inter relationships between the various battery engineering fields that are required to understand the battery as an Energy Storage System It gives great insights for readers ranging from engineers to sales marketing management leadership investors and government officials Adds a brief history of battery technology and its evolution to current technologies Expands and updates the chemistry to include the latest types Discusses thermal runaway and cascading failure mitigation technologies Expands and updates the descriptions of the battery module and pack components and systems Adds description of the manufacturing processes for cells modules and packs Introduces and discusses new topics such as battery as a service cell to pack and cell to chassis designs and wireless BMS **Holt Chemistry** Ralph Thomas Myers, 2004 *Study Guide and Solutions Manual to accompany Basic Concepts of Chemistry, 9th Edition* Leo J. Malone, 2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry Basic Chemistry Steven S. Zumdahl, 2004 Description Not Yet Available **Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems **Structure & Function of the**

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