

Concentration and Molarity (M) table

Name: _____

Mass concentration (mg/L)

Using the information on the concentration table, calculate the molarity

Mass of compound	Mass of solvent	Mass of compound	Mass of compound	Mass of solvent	Mass of compound
100	75	0.007	0.005	75	100
200	200	2.15	200	0.15	1.4
1.5	20	1.5	1.5	6.4	0.40
20	20	1.5	54.4	6.4	6.5

Concentration (molarity) and Molarity (M) table

1. **0.007** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
2. **0.005** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
3. **2.15** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
4. **200** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
5. **0.15** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
6. **1.4** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
7. **1.5** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
8. **54.4** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
9. **6.4** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.
10. **6.5** mg/L is a very low concentration. It is a very low concentration. It is a very low concentration.

Concept Review Concentration And Molarity Answer Key

James R. Braun, Sherman



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Braun, Sherman, 1999 *College Chemistry* Henry Fuller Holtzclaw, William R. Robinson, William Harrison Nebergall, 1984

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