

Chemistry Stage 6 Preliminary Course Study Notes

(not including Gibbs energy & electron configuration: SPDF)

ALL EQUATIONS: mass in grams, volume in litres, temperature in Kelvins

Converting from °C to K: $K = ^\circ C + 273.15$ $0^\circ C = 273.15K$

Module 1: Properties and Structure of Matter

Properties of Matter

Particle Theory of Matter

All matter is made of tiny particles with empty spaces between them. Different substances are made of different types of particles. As the temperature increase particles move faster. Particles attract each other.

Pure Substances

Element: Consists of one type of atom that cannot be broken down to anything simpler.

Can exist as atoms or molecules eg. oxygen

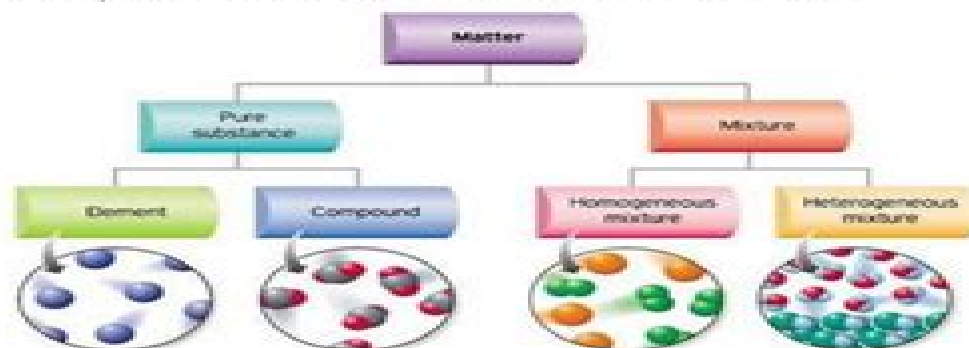
Compound: Consists of two or more types of atoms that are chemically bonded together in a fixed ratio. Can be chemically broken down into simpler compounds or individual elements. Eg. H_2O (water)

Impure Substances (mixtures)

Consist of two or more different elements or compounds physically mixed. Can be physically separated into its components.

Homogeneous Mixture: Are evenly mixed throughout the mixture. This means the same properties and composition of the mixture remains the same throughout it. Eg. NaCl (table salt)

Heterogeneous Mixture: Are NOT evenly mixed, the properties and composition are differing throughout. Eg. A mixture of water & oil.



Calculating Percentage Composition

Gravimetric analysis is used to calculate the percentage composition (what % of each element is present) of a mixture by measuring a change in mass.

$$\% \text{ weight} = \frac{\text{Mass of component in sample (g)}}{\text{Total mass of sample (g)}} \times 100$$

Separation Methods

Separation Method	Description
Sieving	Using solids of different particle sizes. Sieve retains larger particles
Sedimentation & Decantation	Based on density of particles. Large particles separate and 'settle' under the effect of gravity. <div data-bbox="1352 1042 1835 1142"> </div>
Filtration	Based on particle size, mixture is poured through filter paper leaving insoluble residue as the filtrate passes through.
Evaporation (boiling)	Liquids containing different boiling points. Liquid evaporates leaving, dissolved solution.

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Tobias Bleicker



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web stevelinden com classic car appraisals 9 mobrey lane smithtown ny 11787 516 524 4102 stevelinden com e mail steve stevelinden com documents in your possession pertaining to vehicle original bill of sale window sticker receipts etc ownership documents and state title registration etc note any liens