

Edexcel IGCSE Chemistry (4CH0)

The page numbers shown in the right-hand column refer to the **Chemistry for You Students' Book** (ISBN 978-1-4085-0931-7).

In addition there are many printable **Help & Extension Sheets** in the **Kerboodle book** (ISBN 978-1-4085-0918-7)

Paper 1 will assess only the content which is not in **blue, bold**.

Paper 2 will assess all content including content in **blue, bold**.

Section 1: Principles of chemistry

Page number

a) States of matter

Students will be assessed on their ability to:

1.1 understand the arrangement, movement and energy of the particles in each of the three states of matter: solid, liquid and gas

P19

1.2 describe how the interconversion of solids, liquids and gases are achieved and recall the names used for these interconversions

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1.3 describe the changes in arrangement, movement and energy of particles during these interconversions.

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b) Atoms

Students will be assessed on their ability to:

1.4 describe simple experiments leading to the idea of the small size of particles and their movement including:
i dilution of coloured solutions
ii diffusion experiments

P18

1.5 understand the terms atom and molecule

P20-1

1.6 understand the differences between elements, compounds and mixtures

P22-3

1.7 describe techniques for the separation of mixtures, including simple distillation, fractional distillation, filtration, crystallisation and paper chromatography.

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P128, 132, 175, 182

c) Atomic structure

Students will be assessed on their ability to:

1.8 recall that atoms consist of a central nucleus, composed of protons and neutrons, surrounded by electrons, orbiting in shells

P30

1.9 recall the relative mass and relative charge of a proton, neutron and electron

P31

1.10 understand the terms atomic number, mass number, isotopes and relative atomic mass (A_r)

P33, 36

1.11 calculate the relative atomic mass of an element from the relative abundances of its isotopes

P36

1.12 understand that the Periodic Table is an arrangement of elements in order of atomic number

P40

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G Psacharopoulos



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