. Balancing Equations Worksheet ___ Na3PO4 + ___ KOH > ___ NaOH + ___ K3PO4 - NO PO4 Double replacemt reaction 1) ____ MgF₂ + ___ Li₂CO₃ → ___ MgCO₃ + ___ LiF 2) $P_4 + O_2 \rightarrow P_2O_3$ 3) ____ RbNO₃ + ____ BeF₂ → ____ Be(NO₃)₂ + ____ RbF 4) ____ AgNO₃ + ____ Cu → ___ Cu(NO₃)₂ + ____ Ag 5) ____ CF₄ + ___ Br₂ > ___ CBr₄ + ___ F₂ 6) ___ HCN + ___ CuSO₄ → ___ H₂SO₄ + ___ Cu(CN)₂ 7) ____ GaF₃ + ___ Cs -> ___ CsF + ___ Ga 8) ____ BaS + ___ PtF2 -> ____ BaF2 + ___ PtS 9) $N_2 + H_2 \rightarrow NH_3$ 10) ____ NaF + ___ Br₂ -> ___ NaBr + ___ F₂ 11) ____ Pb(OH)₂ + ___ HCl → ___ H₂O + ___ PbCl₂ 12) 13) ____ AlBr₃ + ___ K₂SO₄ → ___ KBr + ___ Al₂(SO₄)₃ $CH_4 + __O_2 \rightarrow __CO_2 + __H_2O$

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