Name: \_\_\_\_\_

Score: \_\_\_\_

## **Net Ionic Equations**

Complete and balance the following equations. Show the ionic and net ionic forms of the equations.

1) K<sub>3</sub>PO<sub>4</sub>(aq.) + AI(NO<sub>3</sub>)<sub>3</sub>(aq.) -

lonic:

Net ionic:

2) Bel<sub>2</sub>(aq.) + Cu<sub>2</sub>SO<sub>4</sub>(aq.)

Ionic:

Net ionic:

3)  $AgNO_3(aq.) + MgI_2(aq.)$ 

lonic:

Net ionic:

4) HBr(aq.) + Pb(ClO<sub>4</sub>)<sub>2</sub>(aq.)

Ionic:

Net ionic:

5)  $K_2(C_2O_4)(aq.) + CaCl_2(aq.)$ 

lonie:

Net ionic:

6)  $Na_2CO_3(aq.)$  +  $FeCl_2(aq.)$ 

lonic:

Net ionic:

7)  $Ni(NO_3)_2(aq.) + NaOH(aq.)$ 

lonic:

Net ionic:

8)  $Pb(NO_3)_2(aq.) + H_2SO_4(aq.)$ 

lonic:

Net ionic:

# **Balancing Net Ionic Equations Wksheet Answer Key**

**M Lipman** 

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