

## STOICHIOMETRY

## Practice Problems

10 problems, 10 minutes each, for practice before the exam.

## SECTION 12.1: CHEMICAL CALCULATIONS

1. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

2. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Molecular formula} = \text{C}_4\text{H}_8\text{O}_4$$

3. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

4. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Molecular formula} = \text{C}_4\text{H}_8\text{O}_4$$

5. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

6. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

7. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Molecular formula} = \text{C}_4\text{H}_8\text{O}_4$$

8. A compound is found to consist of 40.0% oxygen, 53.3% carbon, and 6.7% hydrogen. It has a molar mass of 120.0 g/mol.

$$\text{Molecular formula} = \text{C}_4\text{H}_8\text{O}_4$$

# Bonar Law Chapter 12 Practice Problems Stoichiometry Answers

**Zhe Xu**



**Bonar Law Chapter 12 Practice Problems Stoichiometry Answers:**

## **Bonar Law Chapter 12 Practice Problems Stoichiometry Answers** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has become more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Bonar Law Chapter 12 Practice Problems Stoichiometry Answers**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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