Bendina Basics

ANSWER KEY

Section A: Complete the chart using a periodic table to help you.

Element	Atomic Symbol	Total # of Electrons	# of Valence Electrons	# of Electrons Needed to Gain or Lose to Fill Outer Shells	Oxidation Number
Chlorine	Cl	17	7	Gain 1	1-
Potassium	к	19	1	Lose I	1+
Magnesium	Mg	12	2	Lose 2	2+
Fluorine	F	9	7	Gain I	1-
Aluminum	Al	13	3	Lose 3	3+
Sodium	Na	11	1	Lose 1	1+
Nitrogen	N	14	4	Gain 3	3-
Oxygen	0	8	6	Gain 2	2-
Hydrogen	н	1	1	Gain or Lose 1	1+ or 1-
Carbon	C	6	4:	Gain or Lose 4	4+ or 4-
Lodine	1	53	7	Gain 1	1-

Answer these questions:

- An atom that gains one or more electrons will have a NEGATIVE charge.
- An atom that loses one or more electrons will have a POSTIVE charge.
- An atom that gains or loses one or more electrons is called an ION.
- · A positive ion is called a CATION and a negative ion is called an ANION.

Bonding Basics Review Key

Mark Steyvers

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Table of Contents Bonding Basics Review Key

- 1. Understanding the eBook Bonding Basics Review Key
 - The Rise of Digital Reading Bonding Basics Review Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Bonding Basics Review Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Bonding Basics Review Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Bonding Basics Review Key
 - Personalized Recommendations
 - Bonding Basics Review Key User Reviews and Ratings
 - Bonding Basics Review Key and Bestseller Lists
- 5. Accessing Bonding Basics Review Key Free and Paid eBooks

- Bonding Basics Review Key Public Domain eBooks
- Bonding Basics Review Key eBook Subscription Services
- Bonding Basics Review Key Budget-Friendly Options
- 6. Navigating Bonding Basics Review Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Bonding Basics Review Key Compatibility with Devices
 - Bonding Basics Review Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Bonding Basics Review Key
 - Highlighting and Note-Taking Bonding Basics Review Key
 - Interactive Elements Bonding Basics Review Key
- 8. Staying Engaged with Bonding Basics Review Key
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Bonding Basics Review Key
- 9. Balancing eBooks and Physical Books Bonding Basics Review Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Bonding Basics Review Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Bonding Basics Review Key
 - Setting Reading Goals Bonding Basics Review Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Bonding Basics Review Key
 - Fact-Checking eBook Content of Bonding Basics Review Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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