

◆ — Determining Empirical Formula Worksheet — ◆

1. A compound contains 36.5% Na, 25.4% S, and 38.1% O. Find its empirical formula.
2. Find the empirical formula of a compound with 53.7% iron and 46.3% sulfur.
3. Analysis of a compound sample indicates that it has 1.04 g K, 0.70 g Cr, and 0.86 g O. What is its empirical formula?
4. An acid commonly used in the automotive industry is shown to be 31.6% phosphorous, 3.1% hydrogen, and 63.5% oxygen. Determine the empirical formula of this acid.

Determining Empirical Formulas Answers

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Langford, Dalius Sagatys, Adrian
George**

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A Visual Analogy Guide to Chemistry, 2e Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student General Chemistry I as a Second Language David R. Klein, 2005-03-16 Many students and instructors are overwhelmed by the vast amount of content and concepts presented in General Chemistry Students often emerge from the course with little understanding of chemical concepts and must be retaught in subsequent courses This supplemental text can be paired with Olmsted Williams Brady Spencer or any other General Chemistry title David Klein is a lecturer at Johns Hopkins University where he teaches Organic and General Chemistry He is a dynamic and creative teacher and uses analogy to help students grasp difficult topics Klein s unique informal voice and manner of presentation help students truly master key topics in this course He is also the author of Organic Chemistry as a Second Language response to this book has been phenomenal

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biomedical sciences A new chapter focuses on using spreadsheets and laboratory information management systems Other chapters cover calculations and techniques relevant to reagents chemical reactions properties of gases and solutions pH and buffer preparation spectrophotometry enzyme assays and radioactivity Also included are derivations of some key equations quick reference guides and an index to the practical examples Efiok is with the National Heart Lung and Blood Institute National Institutes of Health Eduok is in the chemistry department at Xavier U of Louisiana c Book News Inc

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