



# Determining Empirical Formula Worksheet



1. A solvent is found to be 50.0% oxygen, 37.5% carbon, and 12.5% hydrogen. What is the empirical formula of this solvent?
2. A particular sugar is determined to have the following composition: 40.0% carbon, 6.7% hydrogen, and 53.5% oxygen. Determine the empirical formula of this sugar molecule.
3. If 4.04 g of nitrogen combine with 11.46 g of oxygen to produce a compound with a molar mass of 108.0 g, what is the molecular formula of this compound?
4. The molar mass of a compound is 92 g. Analysis of the sample indicates that it contains 0.606 g N and 1.390 g O. Find the compound's molecular formula.

# Determining Empirical Formulas Sheet Answer Key

**Peter Atkins, Loretta Jones**



## **Determining Empirical Formulas Sheet Answer Key:**

**The Practice of Chemistry** Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. [http://www.bfwpub.com/pdfs/wink/POCPowerPoint\\_Final.ppt](http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt) 832KB

**Basic Concepts of Chemistry** Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text.

**Chemistry** John Olmsted, Gregory M. Williams, 1994 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee. Imagine if you were as energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans. Now that's real chemical understanding. Olmsted and Williams' Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist. This text helps you understand how chemical phenomena are governed by what happens at the molecular level, apply critical thinking skills to chemical concepts and problems, and master the basic mathematical techniques needed for quantitative reasoning. You'll see the world as chemists do and learn to appreciate the chemical processes all around us. A Fourth Edition with a lot of new perks. Revisions include a new early energy chapter, revised coverage of bonding, expanded coverage of intermolecular forces, and increased coverage of multiple equilibria including polyprotic acids. New pedagogy strengthens students' critical thinking and problem-solving skills. Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills, concepts, equations, and terms. eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text, links between problems and relevant sections in the online text, practice quizzes, the Visual Tutor, Interactive LearningWare problems, and lab demos.

as well as homework management and presentation features for instructors

*Joining of Materials and Structures* Robert W. Messler, 2004-08-24 Joining of Materials and Structures is the first and only complete and highly readable treatment of the options for joining conventional materials and the structures they comprise in conventional and unconventional ways and for joining emerging materials and structures in novel ways Joining by mechanical fasteners integral designed or formed in features adhesives welding brazing soldering thermal spraying and hybrid processes are addressed as processes and technologies as are issues associated with the joining of metals ceramics including cement and concrete glass plastics and composites including wood as well as for the first time anywhere living tissue While focused on materials issues issues related to joint design production processing quality assurance process economics and joint performance in service are not ignored The book is written for engineers from an in training student to a seasoned practitioner by an engineer who chose to teach after years of practice By reading and referring to this book the solutions to joining problems will be within one's grasp

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**Foundations of Physical Science** Tom Hsu PH.D., 2002

**Chemical Principles** Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding

**Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

**Holt Chemistry** Ralph Thomas Myers, 2004

*Technical Drawing Problems* Henry Cecil Spencer, Ivan Leroy Hill, John T. Dygdon, 1980

**Fundamentals of Chemistry** David E. Goldberg, 2001

*Nomography and Empirical Equations* Dale Strohle Davis, 1962

*Water Works Engineering*, 1928

List of Bureau of Mines Publications and Articles ... with Subject and

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