# Balancing Equations Worksheet - Answers

Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number "1".

- 1 Na<sub>3</sub>PO<sub>4</sub> + 3 KOH → 3 NaOH + 1 K<sub>3</sub>PO<sub>4</sub>
- 2) 1 MgF₂ + 1 Li₂CO₃ → 1 MgCO₃ + 2 LiF
- 3) 1 P<sub>4</sub> + 3 O<sub>2</sub> → 2 P<sub>2</sub>O<sub>3</sub>
- 4) 2 RbNO<sub>3</sub> + 1 BeF<sub>2</sub> → 1 Be(NO<sub>3</sub>)<sub>2</sub> + 2 RbF
- 5) 2 AgNO<sub>3</sub> + 1 Cu → 1 Cu(NO<sub>3</sub>)<sub>2</sub> + 2 Ag
- 6) 1 CF<sub>4</sub> + 2 Br<sub>2</sub> → 1 CBr<sub>4</sub> + 2 F<sub>2</sub>
- 7) 2 HCN + 1 CuSO<sub>4</sub> → 1 H<sub>2</sub>SO<sub>4</sub> + 1 Cu(CN)<sub>2</sub>
- 8) 1 GaF<sub>3</sub> + 3 Cs → 3 CsF + 1 Ga
- 9) 1 BaS + 1 PtF<sub>2</sub> → 1 BaF<sub>2</sub> + 1 PtS
- 10) 1 N<sub>2</sub> + 3 H<sub>2</sub> → 2 NH<sub>3</sub>
- 11) 2 NaF + 1 Br<sub>2</sub> → 2 NaBr + 1 F<sub>2</sub>
- 12) 1 Pb(OH)<sub>2</sub> + 2 HCl → 2 H<sub>2</sub>O + 1 PbCl<sub>2</sub>
- 13) 2 AlBr<sub>3</sub> + 3 K<sub>2</sub>SO<sub>4</sub> → 6 KBr + 1 Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>
- 14) 1 CH<sub>4</sub> + 2 O<sub>2</sub> → 1 CO<sub>2</sub> + 2 H<sub>2</sub>O
- 15) 2 Na<sub>3</sub>PO<sub>4</sub> + 3 CaCl<sub>2</sub> → 6 NaCl + 1 Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>
- 16) 2 K + 1 Cl<sub>2</sub> → 2 KCl
- 17) 2 Al + 6 HCl → 3 H<sub>2</sub> + 2 AlCl<sub>3</sub>
- 18) 1 N₂ + 3 F₂ → 2 NF₃
- 19) 1 SO₂ + 2 Li₂Se → 1 SSe₂ + 2 Li₂O
- 20) 2 NH<sub>3</sub> + 1 H<sub>2</sub>SO<sub>4</sub> → 1 (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>

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**RJ Alexander** 

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# **Table of Contents Cavalcade Publishing Answers**

- 1. Understanding the eBook Cavalcade Publishing Answers
  - The Rise of Digital Reading Cavalcade Publishing Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Cavalcade Publishing Answers
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Cavalcade Publishing Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Cavalcade Publishing Answers
  - Personalized Recommendations
  - Cavalcade Publishing Answers User Reviews and Ratings
  - Cavalcade Publishing Answers and Bestseller Lists

- 5. Accessing Cavalcade Publishing Answers Free and Paid eBooks
  - Cavalcade Publishing Answers Public Domain eBooks
  - Cavalcade Publishing Answers eBook Subscription Services
  - Cavalcade Publishing Answers Budget-Friendly Options
- 6. Navigating Cavalcade Publishing Answers eBook Formats
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  - Cavalcade Publishing Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Cavalcade Publishing Answers
  - Highlighting and Note-Taking Cavalcade Publishing Answers
  - Interactive Elements Cavalcade Publishing Answers
- 8. Staying Engaged with Cavalcade Publishing Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Cavalcade Publishing Answers
- 9. Balancing eBooks and Physical Books Cavalcade Publishing Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Cavalcade Publishing Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Cavalcade Publishing Answers
  - Setting Reading Goals Cavalcade Publishing Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Cavalcade Publishing Answers
  - Fact-Checking eBook Content of Cavalcade Publishing Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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