

## CBSE Class 12 Chemistry Sample Paper Set 8

CHEMISTRY  
CLASS-XII

Time Allotted: 3 Hrs

Maximum Marks: 70

## General Instructions:

1. All questions are compulsory.
2. Question No. 1-8 are very short answer questions and carry 1 mark each.
3. Question No. 9-18 are short answer questions and carry 2 marks each.
4. Question No. 19-27 are also short answer questions and carry 3 marks each.
5. Question No. 28-30 are long answer questions and carry 5 marks each.
6. Use log tables if necessary, use of calculators is not allowed.

Q.1 Bond enthalpy of fluorine is lower than that of chlorine, why?

Q.2 Give the IUPAC name of the following compound:



Q.3 On increasing temperature, activation energy of a reaction decreases, why?

Q.4 Which of the following is most effective electrolyte in the coagulative of  $\text{AgI}/\text{Ag}+\text{sol}$ ?  
 $\text{K}_2\text{SO}_4$ ,  $\text{MgCl}_2$ ,  $\text{K}_4[\text{Fe}(\text{CN})_6]$

Q.5 Write the reaction when glucose is heated with excess of HI.

Q.6 Which Xe compound has distorted octahedral shape?

Q.7 What is the denticity of co-ordination compound used for the treatment of lead Poisoning?

Q.8 An alkoxide is a stronger base than hydroxide ion. Justify.

Q.9 (a) State the law which helps to determine the limiting molar conductivity of weak electrolyte.

(b) Calculate limiting molar conductivity of  $\text{CaSO}_4$  (limiting molar conductivity of calcium and sulphate ions are 119.0 and 160.0  $\text{S cm}^2 \text{mol}^{-1}$  respectively).

Q.10 Rate constant K for first order reaction has been found to be  $2.54 \times 10^{-3} \text{ sec}^{-1}$ . Calculate its three-fourth life.

OR

A first order gas reaction  $\text{A}_{2(\text{g})}\text{B}_{2(\text{g})} \rightarrow 2\text{A}_{(\text{g})} + 2\text{B}_{(\text{g})}$  at the temperature  $400^\circ\text{C}$  has the rate Constant  $K = 2.0 \times 10^{-4} \text{ sec}^{-1}$ . What percentage of  $\text{A}_2\text{B}_2$  is decomposed on heating for 900 seconds.

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*Annual Report India. Department of Elementary Education & Literacy, 2003*

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