

CHM 001 Test Questions

Leach's Chapter 2/Three

- What is the vapor density of R250 4 (R=1,S=32,O=16)(a) 94 (b)98(c)48(d)99
- The atomic number is also called
 - no. of protons
 - no. of electrons
 - no. of atoms
 - no. of neutrons
- The Avogadro's number is (a) 6.023×10^{23} (b) 6.03×10^{23} (c) 6.022×10^{23} (d) 6.04×10^{23}
- Find the number of moles of a substance present in 0.3g of a substance sample
 - 0.27 moles
 - 0.26 moles
 - 0.24 moles
 - 0.25 moles
- Molar mass = _____
 - molar concentration/molar concentration
 - molar concentration/molar concentration
 - number of moles $\times$ molar concentration
 - Molar ratio $\times$ Molar Concentration
- Which of these undergoes oxidation
 - Ammonium Chloride
 - Oxygen Gas
 - Aluminium Oxide
 - Phosphorus Soap
- Covalent bonds is found in (a) NaCl (b) Benzene Methyl (c) AlO (d) Oxygen Molecule
- Surface bond is found in (a) SiH_4 (b) $O=O$ (c) $H-H$ (d) $NaCl$
- Electrovalent bond involves _____ while covalent involves _____ (a) Releasing, Sharing (b) Daring, Transfer (c) Transfer, Sharing
- Allotropes occur in the following Except _____ (a) C (b) S (c) Cl (d) P
- _____ is NOT diatomic (a) O (b) Cl (c) H (d) Al
- The formula of Alkane is _____ (a) C_nH_{2n} (b) C_nH_{2n-1} (c) C_nH_{2n-2} (d) C_nH_{2n+2}
- _____ is the most reactive element and why?
- He, Noble Gases (b) O, easily Available (c) F, High Electronegativity (d) Acid, High Electropositivity
- _____ is the maximum number of valence electrons a metallic atom has (a) 4 (b) 3 (c) 2 (d) 1
- A negative has what charge? (a) Positive (b) Negative (c) Electric (d) 2
- _____ is the ability of an element to have same structural formula/molecular formula but different physical properties
- Separated in Organic Chemistry Mixture _____ (a) Partially Filled (b) Completely Filled (c) Selectively Filled (d) Irregularly Filled
- Mythology is Popularly Called (a) Cosmology (b) Science (c) Bio/Of Death (d) Style
- _____ has same MASS number but different atomic numbers (a) Isomer (b) Isobar (c) Allotrope (d) Isoprene
- The Atomic theory was proposed by _____ (a) Prof. Sir Isaac Newton (b) Robert Brown (c) Lord Rutherford (d) John Dalton
- The Major Difference Between "Distillation" And Fractional Distillation Is (a) Wide Boiling Points of Liquids (b) close boiling points (c) Solubility Curve (d) Range Of Boiling point
- A Mixture May Be _____ (a) Molecular Formula (b) Structural Formula (c) Formula (d) Isotope
- _____ Separating Technique involves "Mixture" (a) Magnetism (b) Distillation
- _____ (c) Polysaccharide (d) Centrifugation
- Which Of These is not a type of hybridization (a) sp^3 (b) sp^2 (c) sp (d) sp
- What Is The Valency Of Carbon (a) 4 (b) 3 (c) 5 (d) 2
- Bonding Occurs between _____ of elements (a) Valence bond (b) Valence electrons (c) Valence Atom (d) Valence Molecules
- _____ is saturated (a) C_2H_2 (b) C_2H_4 (c) C_2H_6 (d) C_2H_2
- _____ And _____ Decomposes Besides Water (a) Ethane And Methane (b) Ethane And Propanoic Acid (c) Ethane And Pentane (d) Propane And Butene
- _____ is the breakdown of a compound by the action of heat (a) Baking (b) Cracking (c) Polymerization (d) Distillation
- _____ is not a type of isomerism (a) Functional group (b) Chain (c) Molecular (d) Positional
- There is a variation of _____ and _____ in the homologous (a) Physical and chemical properties (b) Molecular and properties (c) States and structures (d) Boiling And melting points
- The Ability of an element to attract and hold water from the atmosphere is called _____ (a) Deliquescence (b) Coagulant (c) Hygroscopy (d) Desiccation
- Destructive Distillation Of Coal is the heating of coal to very high temperatures in the absence of air to give _____ main products (a) 3 (b) 2 (c) 3 (d) 4

"I Wish You All The Best In Your Post-UTME Examination, I Pray Your Reaction Drive Forward, Grace And mercy In Jesus Name, Amen"

Chm 1tma Questions

Wolfgang Guggemos

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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