

**Chemistry: Unit 10 Solubility Test Review**

1. In a solution, the substance being dissolved is the \_\_\_\_\_.
2. The most commonly used solvent is \_\_\_\_\_.
3. Ethylene glycol (antifreeze) and water are completely miscible. If you mix 20 mL of ethylene glycol with 10 mL of water in a beaker, what would you expect to see?
4. A solution of sodium chloride and water is what type of mixture?
5. The most precise description of water and oil would be to say that these two substances are \_\_\_\_\_.
6. Give several examples of an emulsifying agent?
7. What are the properties of an emulsifying agent? (In other words, what do they do? For what are they used?)
8. A silver colored dental filling is a(s) \_\_\_\_\_ and an iodine solution for cuts is a(n) \_\_\_\_\_.
9. Give two examples of alloys.
10. What are the four factors that affect the rate of dissolution?

11. Some drugs taken by athletes can show up on drug tests many months after the athlete last took them. These drugs are probably \_\_\_\_\_ and \_\_\_\_\_.

Use your solubility curve to answer the next five questions.

12. Between 40°C and 60°C, which compound shows the largest increase in solubility?
13. The solubility of which compound is LEAST affected by temperature?
14. The solubility of  $KNO_3$  increases by approximately \_\_\_\_\_ when the temperature rises from 10°C to 80°C.
15. For 140 g of KI (in 100 g  $H_2O$ ) at 15°C, the solution would be described as being \_\_\_\_\_.
16. Which of the following solutions is unsaturated?  
80 g of KCl (in 100 g of water) at 80°C  
40 g of NaCl (in 100 g of water) at 50°C  
140 g of  $NaNO_3$  (in 100 g of water) at 40°C  
40 g of  $H_2SO_4$  (in 100 g of water) at 40°C
17. A solution of HCl is 0.380 M. Determine the mass of HCl needed to make 150 mL of solution.
18. A 2.50 M solution of  $Na_2CO_3$  contains 87 g of solute. Find the volume of the solution.

# Chemistry Unit Test Review Answers

**Herbert Drapkin**



## Chemistry Unit Test Review Answers:

**Phlebotomy Exam Review, Enhanced Edition** Ruth E. McCall, 2020-06-19 Phlebotomy Exam Review Enhanced Seventh Edition thoroughly prepares students for any of the national certification exams in phlebotomy *General Chemistry Study Guide Sixth Edition* Darrell D. Ebbing, 1999 **Learning to Assess** Christopher DeLuca, Jill Willis, Bronwen Cowie, Christine Harrison, Andrew Coombs, 2024-01-01 This book presents a new framework for how teachers develop their assessment capacity based on a multi year study conducted in four countries Australia Canada England and New Zealand which focused on student teacher learning in assessment throughout their initial teacher education programs It examines how teacher learning is shaped by the complex dynamics of assessment capacity within larger teacher education contexts The framework proposed here identifies four domains involved in cultivating assessment capacity and characterizes assessment learning as always integrating cognitive philosophical and moral dimensions with assessment's social emotional and physical dimensions while recognizing that each capacity is continually shaped by the learning context The book draws on the survey of teacher education programs in each of the four focal countries and data from student teachers to shed light on how the various pedagogies program structures and policies encountered provide beginning teachers with codes for classifying and framing assessment capacity and form a template for developing this capacity throughout their careers Offering suggestions for future research and teacher education practice the book concludes with an outlook on future steps to cultivate teachers assessment capacity *Study Guide* Joan I. Senyk, Darrell D. Ebbing, 1984 **Developing Successful Schools** Stephen P. Gordon, 2022-07-21 This book identifies and emphasizes the need for a holistic approach to school improvement when it comes to both the development of the whole child and the relationships among student family and community development In recent years the emphasis in PK 12 education in the United States has been on the measurement of student and school performance by high stakes achievement tests This emphasis has resulted in a narrowed curriculum emphasizing lower level cognitive learning with little attention paid to the moral social and creative development of students families and communities This book argues that PK 12 education needs to shift its focus to holistic qualities of the successful school qualities that reflect a moral rather than a technical approach to education while also improving students academic performance *General Chemistry* Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996 **Instructional Tests in Chemistry** Earl Rouse Glenn, Louis E. Welton, 1929 *The Indiana Teacher*, 1936 **Improving High School Students' Performance in Chemistry with a Hands-on Approach** Mary L. Fredell, 1998 *The Software Encyclopedia 2001*, 2001 *Teacher's Manual and Resource Guide for Exploring the Sciences* Herbert Drapkin, 1964 **The Science Teacher**, 1988-04 *Gage Canadian GED Study Guide Series* Kathy Austin, 1998 5 Steps to a 5: 500 AP Chemistry Questions to Know by Test Day, Fourth Edition Mina Lebitz, 2022-02-21 The only study guide you'll need for the AP Chemistry test revised and updated now with a 20 question Diagnostic Quiz Confidence is key when taking any exam and it will come easier if you

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