

Date ____/____/____

CH:2 Solutions

• Notes

- Solutions are homogeneous mixture of two or more than two components.
- The component present in large extent in solution is called solvent whereas the one or more component present in the solution other than solvent are called solutes.

Types of Solutions.

	<u>Solute</u>	<u>Solvent</u>
<u>Gaseous Solⁿ</u> :	Gas	Gas → Mixture of O_2 and N_2 gas
	Liquid	Gas → Chloroform mixed with N_2 gas
	Solid	Gas → Carbon in N_2 gas
<u>Liquid Solⁿ</u> :	Gas	liquid → Oxygen dissolved in water
	Liquid	liquid → Ethanol dissolved in water
	Solid	liquid → Glucose dissolved in water
<u>Solid Solⁿ</u> :	Gas	Solid → Sol ⁿ of H_2 in palladium
	Liquid	Solid → Amalgam of mercury in Ni
	Solid	Solid → Copper dissolved in Ag

Expressing Concentration of Solution.i) Mole fraction :

$$x_A = \frac{n_A}{n_A + n_B}$$

Class Xii Chemistry Notes

WJ Hussar



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