

- Specific Nature** - Urinase catalyses the hydrolysis of urea and not methyl urea, so these are specific in nature.
- Optimum Temperature** - It is active at 20-30°C
- pH of medium** - It is about 7, for pepsin 1.5-2.5, for trypsin 7.5-8.5
- Concentration** - Dilute solutions are more effective
- Amount of enzyme** - Very small amount can accelerate the reaction
- Enzyme inhibitors** - These compound inhibit the enzyme action, with the help of such compounds, the reaction can be controlled.

HORMONES These are the chemical substances which are produced by endocrine (ductless) glands in the body. Hormones act as chemical messengers.

Some examples of ductless (endocrine) glands are thyroid, pituitary, adrenal, pancreas, testes and ovaries.

Hormones are divided into three types:

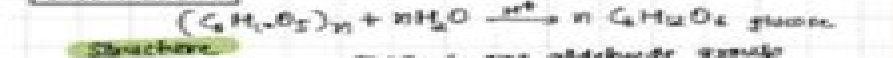
- 1) Steroids
- 2) Peptides
- 3) Amines

Name of endocrine gland	Hormones secreted
Testis	Testosterone
Ovary	Estrogen
Adrenal	Adrenaline
Thyroid glands	Thyroxine
Pituitary gland	FSH
Pancreas	Insulin

Carbohydrates These are optically active polyhydroxy aldehydes / ketones or the substance which give these on hydrolysis are called carbohydrates.

- Classification on the basis of Hydrolysis**
- Mono-saccharides** - can not be hydrolyzed further e.g. glucose, fructose.
 - Oligo-saccharides** - give 2-10 molecules of mono-saccharides e.g. sucrose, maltose, raffinose.
 - Polysaccharides** - give large no. of mono-saccharides e.g. starch, cellulose.

Preparation of Glucose:



Structure

CHO ← one aldehyde group

(CHOH)₄ ← four 2° alcohol

CH₂OH ← one 1° alcohol

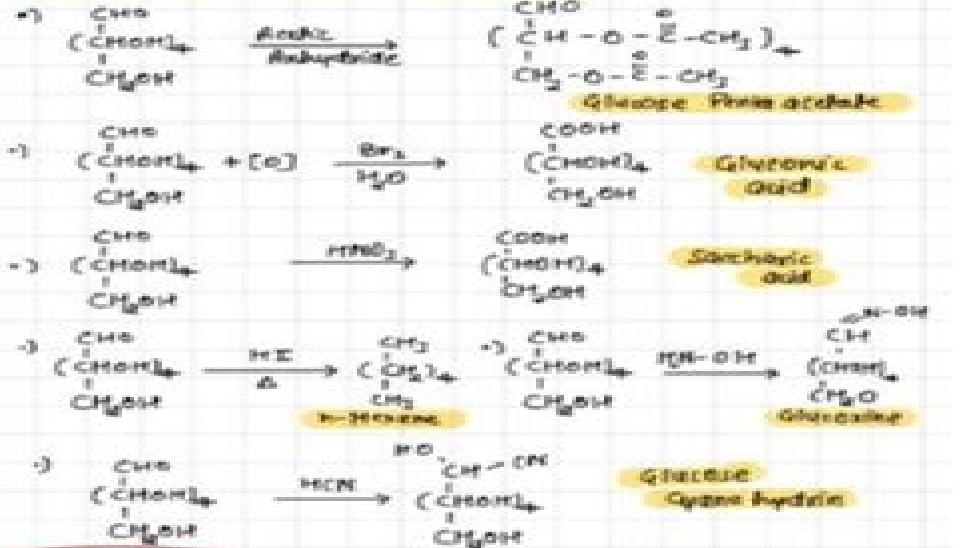


Str. of fructose: { The C, 6 known as anomeric carbon and the other carbons are called anomers }



- Reducing Sugar**
- Free Aldehydic or Ketonic group
 - Reduce Fehling Solⁿ or Tollen's Reagent
 - e.g. Maltose & Fructose
- Non Reducing Sugar**
- Do not have any free aldehydic or ketonic group.
 - Do not reduce Fehling's Reagent and Fehling Solⁿ
 - e.g. Sucrose

Chemical Properties of Glucose



Mutarotation when either of the two forms of glucose is dissolved in water, there is a spontaneous change in specific rotation till the equilibrium value of +52.5°. This is known as mutarotation.

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1. Understanding the eBook Class1chapterwise Chemistry Notes
 - The Rise of Digital Reading Class1chapterwise Chemistry Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Class1chapterwise Chemistry Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Class1chapterwise Chemistry Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Class1chapterwise Chemistry Notes
 - Personalized Recommendations
 - Class1chapterwise Chemistry Notes User Reviews and Ratings
 - Class1chapterwise Chemistry Notes and Bestseller Lists
5. Accessing Class1chapterwise Chemistry Notes Free and Paid eBooks
 - Class1chapterwise Chemistry Notes Public Domain eBooks
 - Class1chapterwise Chemistry Notes eBook Subscription Services
 - Class1chapterwise Chemistry Notes Budget-Friendly Options
6. Navigating Class1chapterwise Chemistry Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Class1chapterwise Chemistry Notes Compatibility with Devices
 - Class1chapterwise Chemistry Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Class1chapterwise Chemistry Notes
 - Highlighting and Note-Taking Class1chapterwise Chemistry Notes
 - Interactive Elements Class1chapterwise Chemistry Notes
8. Staying Engaged with Class1chapterwise Chemistry Notes

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Class1chapterwise Chemistry Notes
9. Balancing eBooks and Physical Books Class1chapterwise Chemistry Notes
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Class1chapterwise Chemistry Notes
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Class1chapterwise Chemistry Notes
- Setting Reading Goals Class1chapterwise Chemistry Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Class1chapterwise Chemistry Notes
- Fact-Checking eBook Content of Class1chapterwise Chemistry Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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