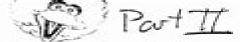
CLASSIFYING CHEMICAL REACTIONS

H,SO,

14. SO₃ + H₂O →



Classify the following reactions as synthesis, decomposition, single replacement or double replacement. AACCLERS

1. $2KClO_3 \rightarrow 2KCl + 3O_2$ Decomposition 2. $HCl + NaOH \rightarrow NaCl + H_2O$ Double Replacement 3. $Mg + 2HCl \rightarrow MgCl_2 + H_2$ Single Replacement 4. $2H_2 + O_2 \rightarrow 2H_2O$ Synthesis 5. $2Al + 3NlBr_2 \rightarrow 2AlBr_3 + 3Nl$ Single Replacement 6. $4Al + 3O_2 \rightarrow 2Al_2O_3$ Synthesis 7. $2NaCl \rightarrow 2Na + Cl_2$ Decomposition 8. $CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$ Single Replacement 9. $AgNO_3 + KCl \rightarrow AgCl + KNO_3$ Double Replacement 10. $N_2 + 3H_2 \rightarrow 2NH_3$ Synthesis 11. $2H_2O_2 \rightarrow 2H_2O + O_2$ Decomposition 12. $(NH_2)_2SO_4 + Bac(NO_3)_2 \rightarrow BasO_4 + 2NH_4NO_3$ Double Replacement 13. $Mgl_2 + Br_2 \rightarrow MgBr_2 + l_2$ Single Replacement	repidcemeni.	ANSW	ERS/
3. Mg + 2HCI → MgCl2 + H2 Single Replacement. 4. 2H2 + O2 → 2H2O Synthesis 5. 2AI + 3NIBr2 → 2AIBr3 + 3NI Single Replacement 6. 4AI + 3O2 → 2AI2O3 Synthesis 7. 2NaCI → 2Na + Cl2 Decomposition 8. CaCl2 + F2 → CaF2 + Cl2 Single Replacement 9. AgNO3 + KCI → AgCI + KNO3 Double Replacement 10. N2 + 3H2 → 2NH3 Synthesis 11. 2H2O2 → 2H2O + O2 Decomposition 12. (NH2)2SO4 + Ba(NO3)2 → BaSO4 + 2NH4NO3 Double Replacement	1. 2KCIO ₃ → 2KCI + 3O ₂	De con	~position
4. $2H_2 + O_2 \rightarrow 2H_2O$ Synthesis 5. $2AI + 3NIBI_2 \rightarrow 2AIBI_3 + 3NI$ Single Replacement 6. $4AI + 3O_2 \rightarrow 2AI_2O_3$ Synthesis 7. $2NaCI \rightarrow 2Na + CI_2$ By composition 8. $CaCI_2 + F_2 \rightarrow CaF_2 + CI_2$ 9. $AgNO_3 + KCI \rightarrow AgCI + KNO_3$ Double Replacement 10. $N_2 + 3H_2 \rightarrow 2NH_3$ Synthesis 11. $2H_2O_2 \rightarrow 2H_2O + O_2$ Decomposition 12. $(NH_2)_2SO_4 + Ba(NO_3)_2 \rightarrow BaSO_4 + 2NH_3NO_3$ Double Replacement	2. HCl + NaOH → NaCl + H₂O	Double	Replacement
5. 2AI + 3NIBr ₂ → 2AIBr ₃ + 3NI Single Replacement 6. 4AI + 3O ₂ → 2AI ₂ O ₃ Synthesis 7. 2NOCI → 2NO + CI ₂ Decomposition 8. COCI ₂ + F ₂ → COF ₂ + CI ₂ Single Replacement 9. AGNO ₃ + KCI → AGCI + KNO ₃ Double Replacement 10. N ₂ + 3H ₂ → 2NH ₃ Synthesis 11. 2H ₂ O ₂ → 2H ₂ O + O ₂ Decomposition 12. (NH ₂ , SO ₄ + BO(NO ₃) ₂ → BOSO ₄ + 2NH ₄ NO ₃ Double Replacement	3. Mg + 2HCl → MgCl ₂ + H ₂	Single	Replacement.
6. 4AI + 30, - 2AI2O, 7. 2NacI - 2Na + CI2 8. CacI2 + F2 - CaF2 + CI2 9. AgNO3 + KCI - AgCI + KNO3 10. N2 + 3H2 - 2NH3 11. 2H2O2 - 2H2O + O2 12. (NH2)2SO4 + Ba(NO3)2 - BaSO4 + 2NH4NO3 Double Replacement Decomposition 12. (NH2)2SO4 + Ba(NO3)2 - BaSO4 + 2NH4NO3 Double Replacement Decomposition	4. 2H ₂ + O ₂ → 2H ₂ O	Synth	esis
7. 2NaCl - 2Na + Cl2 Be composition 8. CaCl2 + F2 - CaF2 + Cl2 9. AgNO3 + KCl - AgCl + KNO3 10. N2 + 3H2 - 2NH3 11. 2H2O2 - 2H2O + O2 De composition De composition De composition De composition De composition	5. 2AI + 3NIBr₂ → 2AIBr₃ + 3NI	Single	Replacement
8. Cacl ₂ + F ₂ → CaF ₂ + Cl ₂ Single Replacement 9. AgNO ₃ + KCl → AgCl + KNO ₃ Double Replacement 10. N ₂ + 3H ₂ → 2NH ₃ Synthesis 11. 2H ₂ O ₂ → 2H ₂ O + O ₂ Decomposition 12. (NH ₂) ₂ SO ₄ + Ba(NO ₃) ₂ → BaSO ₄ + 2NH ₄ NO ₃ Double Replacement	6. 4AI + 3O₂ → 2AI₂O₃	Synth	e515
9. AgNO3 + KCI - AgCI + KNO3 Double Replacement 10. N2 + 3H2 - 2NH3 Synthesis 11. 2H2O2 - 2H2O + O2 Decomposition 12. (NH2)2SO4 + Ba(NO3)2 - BaSO4 + 2NH2NO3 Double Replacement	7. 2NaCl → 2Na + Cl ₂	Decom	position
10. N ₂ + 3H ₂ → 2NH ₃ Synthesis 11. 2H ₂ O ₂ → 2H ₂ O + O ₂ Decomposition 12. (NH ₂) ₂ SO ₄ + Ba(NO ₃) ₂ → BaSO ₄ + 2NH ₄ NO ₅ Double Replacement	8. $CaCl_2 + F_2 \rightarrow CaF_2 + Cl_2$	Single	Replacement
11. 2H2O2 - 2H2O + O2 De composition 12. (NH2)2SO4 + BO(NO3)2 - BOSO4 + 2NH4NO, Double Replacement	9. AgNO ₃ + KCI → AgCI + KNO ₃	Double	Replacement
12. (NH.)250, + Ba(NO), - Baso, + 2NH, NO, Double Replacement	10. N ₂ + 3H ₂ → 2NH ₃	Synth	resis
	11. 2H ₂ O ₂ → 2H ₂ O + O ₂	Decomp	position
13. Mgl2 + Br2 - MgBr2 + 12 Single Replacement	12. (NH ₄) ₂ SO ₄ + Ba(NO ₃) ₂ → BaSO ₄ +	+ 2NH4NO, DO	uble Replacement
	13. Mgl ₂ + Br ₂ → MgBr ₂ + l ₂	Single	Replacement

Synthesis

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JG Myers

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