

# BSC Computer Science



## 2nd Semester

### V. V Important

### Previous Questions Paper



## आएंगे ये प्रश्न



Structural, Bonding and Molecular Properties of Organic Molecules

Organic chemistry = the study of carbon compounds.

Organic compounds are divided into two main classes:

- 1. Aliphatic compounds (open chain compounds)
- 2. Aromatic compounds (closed ring compounds)

Q. Define, Bonding:

Ans. Bonding is the process of formation of chemical bonds between atoms.

Q. Define, Valency:

Ans. Valency is the combining capacity of an element.

Q. Define, Atomicity:

Ans. Atomicity is the number of atoms present in a molecule.

Q. Define, Molecular weight:

Ans. Molecular weight is the sum of atomic weights of all the atoms present in a molecule.

Q. Define, Empirical formula:

Ans. Empirical formula is the simplest formula representing the relative number and kinds of atoms in a compound.

Q. Define, Molecular formula:

Ans. Molecular formula is the formula which represents the actual number and kinds of atoms in a molecule.

Q. Define, Structural formula:

Ans. Structural formula is the formula which represents the arrangement of atoms in a molecule.

Q. Define, Lewis structure:

Ans. Lewis structure is the structural formula in which the valence electrons are represented by dots.

Q. Define, Hybridization:

Ans. Hybridization is the process of intermingling of atomic orbitals to form new hybrid orbitals.

Q. Define, Hybrid orbital:

Ans. Hybrid orbital is the orbital formed by the hybridization of atomic orbitals.

Q. Define, Hybridization of sp<sup>3</sup>:

Ans. Hybridization of sp<sup>3</sup> is the process of intermingling of one s orbital and three p orbitals to form four sp<sup>3</sup> hybrid orbitals.

Q. Define, Hybridization of sp<sup>2</sup>:

Ans. Hybridization of sp<sup>2</sup> is the process of intermingling of one s orbital and two p orbitals to form three sp<sup>2</sup> hybrid orbitals.

Q. Define, Hybridization of sp:

Ans. Hybridization of sp is the process of intermingling of one s orbital and one p orbital to form two sp hybrid orbitals.

Q. Define, Hybridization of d:

Ans. Hybridization of d is the process of intermingling of one d orbital and one s orbital to form two d hybrid orbitals.

Q. Define, Hybridization of p:

Ans. Hybridization of p is the process of intermingling of one p orbital and one s orbital to form two p hybrid orbitals.

Q. Define, Hybridization of s:

Ans. Hybridization of s is the process of intermingling of one s orbital and one p orbital to form two s hybrid orbitals.

Q. Define, Hybridization of f:

Ans. Hybridization of f is the process of intermingling of one f orbital and one s orbital to form two f hybrid orbitals.

Q. Define, Hybridization of g:

Ans. Hybridization of g is the process of intermingling of one g orbital and one s orbital to form two g hybrid orbitals.

Q. Define, Hybridization of h:

Ans. Hybridization of h is the process of intermingling of one h orbital and one s orbital to form two h hybrid orbitals.

Q. Define, Hybridization of i:

Ans. Hybridization of i is the process of intermingling of one i orbital and one s orbital to form two i hybrid orbitals.

Q. Define, Hybridization of j:

Ans. Hybridization of j is the process of intermingling of one j orbital and one s orbital to form two j hybrid orbitals.

Q. Define, Hybridization of k:

Ans. Hybridization of k is the process of intermingling of one k orbital and one s orbital to form two k hybrid orbitals.

Q. Define, Hybridization of l:

Ans. Hybridization of l is the process of intermingling of one l orbital and one s orbital to form two l hybrid orbitals.

Q. Define, Hybridization of m:

Ans. Hybridization of m is the process of intermingling of one m orbital and one s orbital to form two m hybrid orbitals.

Q. Define, Hybridization of n:

Ans. Hybridization of n is the process of intermingling of one n orbital and one s orbital to form two n hybrid orbitals.

Q. Define, Hybridization of o:

Ans. Hybridization of o is the process of intermingling of one o orbital and one s orbital to form two o hybrid orbitals.

Q. Define, Hybridization of p:

Ans. Hybridization of p is the process of intermingling of one p orbital and one s orbital to form two p hybrid orbitals.

Q. Define, Hybridization of q:

Ans. Hybridization of q is the process of intermingling of one q orbital and one s orbital to form two q hybrid orbitals.

Q. Define, Hybridization of r:

Ans. Hybridization of r is the process of intermingling of one r orbital and one s orbital to form two r hybrid orbitals.

Q. Define, Hybridization of s:

Ans. Hybridization of s is the process of intermingling of one s orbital and one p orbital to form two s hybrid orbitals.

Q. Define, Hybridization of t:

Ans. Hybridization of t is the process of intermingling of one t orbital and one s orbital to form two t hybrid orbitals.

Q. Define, Hybridization of u:

Ans. Hybridization of u is the process of intermingling of one u orbital and one s orbital to form two u hybrid orbitals.

Q. Define, Hybridization of v:

Ans. Hybridization of v is the process of intermingling of one v orbital and one s orbital to form two v hybrid orbitals.

Q. Define, Hybridization of w:

Ans. Hybridization of w is the process of intermingling of one w orbital and one s orbital to form two w hybrid orbitals.

Q. Define, Hybridization of x:

Ans. Hybridization of x is the process of intermingling of one x orbital and one s orbital to form two x hybrid orbitals.

Q. Define, Hybridization of y:

Ans. Hybridization of y is the process of intermingling of one y orbital and one s orbital to form two y hybrid orbitals.

Q. Define, Hybridization of z:

Ans. Hybridization of z is the process of intermingling of one z orbital and one s orbital to form two z hybrid orbitals.

# Computer Practical 2question Paper Bsc

**Annelies Wilder-Smith**



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