

# Chemical Reactions

## Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- |               |   |
|---------------|---|
| <u>yes</u>    | 1. A tomato smells rotten.  |
| <u>no</u>     | 2. A drinking glass breaks into smaller pieces.                     |
| <u>no</u>     | 3. A piece of ice melts.  |
| <u>yes</u>    | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u>    | 5. Candle wax burns.  |
| <u>yes/no</u> | 6. Molten candle wax solidifies.                                    |
| <u>yes</u>    | 7. Green leaves turn yellow and red as the seasons change.          |
| <u>yes</u>    | 8. Baking powder produces a gas that makes a cake rise.             |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

# Concept Review Describing Chemical Reactions Answer Key

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## **Table of Contents Concept Review Describing Chemical Reactions Answer Key**

1. Understanding the eBook Concept Review Describing Chemical Reactions Answer Key
  - The Rise of Digital Reading Concept Review Describing Chemical Reactions Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Concept Review Describing Chemical Reactions Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Concept Review Describing Chemical Reactions Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Concept Review Describing Chemical Reactions Answer Key
  - Personalized Recommendations
  - Concept Review Describing Chemical Reactions Answer Key User Reviews and Ratings
  - Concept Review Describing Chemical Reactions Answer Key and Bestseller Lists
5. Accessing Concept Review Describing Chemical Reactions Answer Key Free and Paid eBooks

- Concept Review Describing Chemical Reactions Answer Key Public Domain eBooks
  - Concept Review Describing Chemical Reactions Answer Key eBook Subscription Services
  - Concept Review Describing Chemical Reactions Answer Key Budget-Friendly Options
6. Navigating Concept Review Describing Chemical Reactions Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Concept Review Describing Chemical Reactions Answer Key Compatibility with Devices
  - Concept Review Describing Chemical Reactions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Concept Review Describing Chemical Reactions Answer Key
  - Highlighting and Note-Taking Concept Review Describing Chemical Reactions Answer Key
  - Interactive Elements Concept Review Describing Chemical Reactions Answer Key
8. Staying Engaged with Concept Review Describing Chemical Reactions Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Concept Review Describing Chemical Reactions Answer Key
9. Balancing eBooks and Physical Books Concept Review Describing Chemical Reactions Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Concept Review Describing Chemical Reactions Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Concept Review Describing Chemical Reactions Answer Key
- Setting Reading Goals Concept Review Describing Chemical Reactions Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Concept Review Describing Chemical Reactions Answer Key
- Fact-Checking eBook Content of Concept Review Describing Chemical Reactions Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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