

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Table of Contents Concept Review Holt Chemistry Answers Pg 123

1. Understanding the eBook Concept Review Holt Chemistry Answers Pg 123
 - The Rise of Digital Reading Concept Review Holt Chemistry Answers Pg 123
 - Advantages of eBooks Over Traditional Books
2. Identifying Concept Review Holt Chemistry Answers Pg 123
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Concept Review Holt Chemistry Answers Pg 123
 - User-Friendly Interface
4. Exploring eBook Recommendations from Concept Review Holt Chemistry Answers Pg 123
 - Personalized Recommendations
 - Concept Review Holt Chemistry Answers Pg 123 User Reviews and Ratings
 - Concept Review Holt Chemistry Answers Pg 123 and Bestseller Lists
5. Accessing Concept Review Holt Chemistry Answers Pg 123 Free and Paid eBooks
 - Concept Review Holt Chemistry Answers Pg 123 Public Domain eBooks
 - Concept Review Holt Chemistry Answers Pg 123 eBook Subscription Services
 - Concept Review Holt Chemistry Answers Pg 123 Budget-Friendly Options
6. Navigating Concept Review Holt Chemistry Answers Pg 123 eBook Formats

-
- ePub, PDF, MOBI, and More
 - Concept Review Holt Chemistry Answers Pg 123 Compatibility with Devices
 - Concept Review Holt Chemistry Answers Pg 123 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Concept Review Holt Chemistry Answers Pg 123
 - Highlighting and Note-Taking Concept Review Holt Chemistry Answers Pg 123
 - Interactive Elements Concept Review Holt Chemistry Answers Pg 123
 8. Staying Engaged with Concept Review Holt Chemistry Answers Pg 123
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Concept Review Holt Chemistry Answers Pg 123
 9. Balancing eBooks and Physical Books Concept Review Holt Chemistry Answers Pg 123
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Concept Review Holt Chemistry Answers Pg 123
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Concept Review Holt Chemistry Answers Pg 123
 - Setting Reading Goals Concept Review Holt Chemistry Answers Pg 123
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Concept Review Holt Chemistry Answers Pg 123
 - Fact-Checking eBook Content of Concept Review Holt Chemistry Answers Pg 123
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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