

Math Skills, *continued*

2. Write down the specific heat equation.

$$\text{energy} = cm\Delta T$$

3. Substitute the specific heat, mass, and temperature change values, and solve.

$$\text{energy} = \left(\frac{4,186 \text{ J}}{\cancel{\text{kg}} \cdot \cancel{\text{K}}} \right) \times (200 \cancel{\text{ kg}}) \times (12 \cancel{\text{ K}})$$

$$\text{energy} = 10,000,000 \text{ J} = 1.0 \times 10^4 \text{ kJ}$$

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IM Harris



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