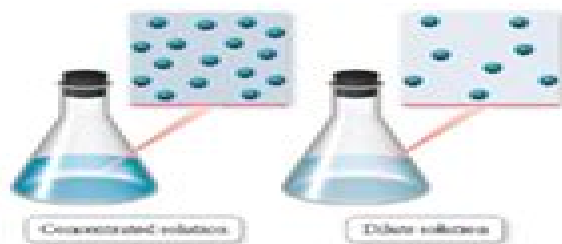


SOLUTION CONCENTRATION

The concentration (c) of a solution refers to the ration of the quantity of solute to the quantity of solution:



Percentage Concentration

The concentration of solutions is sometimes expressed as percentages on consumer products.

| Percent volume/volume % V/V | Percent *weight/volume % W/V | Percent *weight/weight % W/W |
|---|--|---|
| $\% V/V = \frac{\text{volume of solute}}{100 \text{ mL of solution}}$ | $\% W/V = \frac{\text{mass of solute (g)}}{100 \text{ of solution (mL)}}$ | $\% W/W = \frac{\text{mass of solute (g)}}{100 \text{ g of solution}}$ |
| Vinegar is 5% V/V of acetic acid, which means that there is _____ of acetic acid for every _____ of solution. | Hydrogen peroxide is sold as 3% W/V, which means that there is _____ of hydrogen peroxide for every _____ of solution. | Benzoyl peroxide is found in acne cream and is listed as 5% W/W, which means there is _____ of benzoyl peroxide for every _____ of cream. |

*Should really be expressed as mass/volume or mass/mass - but consumer goods still use "weight" on the packaging.

Chemistry Study Guide Solution Concentration Answers

Paul B. Kelter



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Organic Chemistry Study Guide and Solutions Marc Loudon, Jim Parise, 2015-07-01 Parise and Loudon's Study Guide and Solutions Manual offers the following learning aids Links that provide hints for study approaches to problem solving and additional explanations of challenging topics Further Explorations that provide additional depth on key topics Reaction summaries that delve into key mechanisms and stereochemistry Solutions to all the textbook problems Rather than providing just the answer many of the solutions provide detailed explanations of how the problem should be approached Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 **Study Guide and Solutions Manual for**

Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **Introduction to Chemistry, Study Guide** T. R. Dickson, 1999-07-07 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition have been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images **Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and**

Biological Chemistry Katherine J. Denniston, 1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in place of the text Each chapter of the study guide contains the following sections reorganized and expanded set of learning goals concise chapter summary in chapter solved problems list of key terms review problems Preface

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Margolese, David W. Oxtoby, 2004-04

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Holum, 1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples

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