

Term Exam 1 (Practical)

ANSWER ALL THE FOLLOWING QUESTIONS

1) Name 2 properties of a gas. (2 marks)

Indefinite shape & compressible, consisting of moving, fast moving particles

2) Convert 2.1 bar to atm. (2 marks)

$$2.1 \text{ bar} \times \frac{1 \text{ atmosphere}}{1.013 \text{ bar}} = 2.1 \text{ is 1 atm}$$

3) 10.0 g of elemental mercury weighs 22.00 g, what is the density of liquid mercury? (2 marks)

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$10 = \frac{22.00 \text{ g}}{\text{Volume}}$$

$$\text{Volume} = 2.2 \text{ cm}^3$$

4) What are elements in row 6 called? How many elements are in row 6? (2 marks)

See 100

5) Complete the following table. (10 marks)

	Carbon	Chlorine	Argon
Symbol	C	Cl	Ar
Relative atomic mass	12	35.5	40

6) Draw 2 structures of the same element with different numbers of electrons and label each. (4 marks)

Neon

7) Provide the chemical formula. (3 marks)



Chemistry Term Test 1 Answers

RD Boyd



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