Unit 5 - Counting Particles - Objectives

Re	view Concepts	
	 Types of substances Chemical formulas of substances (U 4) 	
1.	State evidence for Avogadro's Hypothesis. Use Avogadro's Hypothesis and experimental data to determine the relative mass of molecules.	
2.	Use experimental data to determine the relative mass of two objects.	
3.	Use experimental data to determine the number of items in a sample without actually counting them.	
4.	Given the chemical formula of a substance, determine the molar mass.	
5.	Given the mass of a substance, determine a. the number of moles of the sample b. the number of atoms or molecules in the sample	

Chemistry Unit 5 Counting Particles Key

Ying Liu

Chemistry Unit 5 Counting Particles Key:

<u>Chemistry</u> Nicholas R. Romano, Philip K. Cameron, 2002 Prepares students for the new standards and the commencement level PS Chemistry Test Challenges with content based multiple choice constructed response and real world thematic questions Stimulates skills based activities in reading writing and lab performance Correlates PS Chemistry key ideas and performance indicators on atomic concepts periodic table moles stoichiometry bonding behavior of matter kinetics organic chemistry oxidation reduction acids bases and salts nuclear chemistry Fosters mastery with practice on three recent tests

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