

## 1.1 Catalysts & activation energy

CATALYST - speeds up reaction

without affecting themselves/  
the products

- allow for lower temps
- ~~higher~~ faster
- reduce activation energy  
(same overall energy change)



ENZYMES - biological catalyst

- have active site, denatured  
by vicarious conditions

GROUPS IN THE PERIODIC TABLE (SC11)  
RATES OF REACTION (SC10)

## 1.2 energy changes in reactions

- reaction happens when  
particles collide enough  
to generate activation  
energy



- BONDS:
- breaking - need energy endothermic
  - making - transfers energy exothermic

## 1.3 exothermic & endothermic (temp inc)

EXO - transfers energy INTO surroundings

ENDO - transfer energy FROM surroundings  
(temp dec)

REACTION PROFILE - demonstrates energy change

- neutralisation and displacement  
ALWAYS ~~endo~~ exothermic

HEAT ENERGY CHANGES IN CHEMICAL  
REACTIONS! (SC10)

# Chemistry Ch 2 answers Study Guide

**Karen Timberlake**



## Chemistry Ch 2 answers Study Guide:

**Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems *Organic Chemistry Study Guide with Solutions Manual* Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer *Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed* Paul A. Bartlett, Judith G. Koch, 1992 *Organic Chemistry Principles and Industrial Practice* Mark M. Green, Harold A. Wittcoff, 2003-09-19 Nylon plexiglas epoxy resin and Elmer's glue dynamite rubber tires and spandex These are a few among the multitude of industrial products produced using the principles of organic chemistry principles that are often taught to students without reference to the commercial and practical importance of the subject The marvelous theoretical principles on which organic chemistry is based are therefore often not fully appreciated by students of this subject Organic chemistry can appear dry meaningless and seemingly irrelevant In this textbook designed to be used in conjunction with classic texts of organic chemistry at the undergraduate level or standing alone for more advanced students two experts M M Green and H A Wittcoff bring the principles and the practice together Written for students and also giving much information that could be used to enhance teaching of the subject the book presented in ten concise chapters combines important industrial processes with the principles of organic chemistry The result is a source of otherwise barely accessible information In addition personal anecdotes from the authors vast experience make this a fascinating and indispensable textbook for everyone wishing to enhance the appreciation of this subject I have never come across such an enticing mix of stories of discovery with basic chemistry Roald Hoffmann Cornell University Simply put this book is a gem The chemistry described is rigorous but the warm humorous and conversational writing style makes the book a joy to read Dasan M Thamattoor Colby College This is a unique fascinating book that bridges organic chemistry principles with chemical industrial applications The story telling style make the reading learning experience extremely enjoyable Qiao Sheng Hu College of Staten Island City University of New York **Study Guide and Solutions Manual** Neil E. Schore, 2002-08-02 *Study Guide to Accompany Calculus for the Management, Life, and Social Sciences* Clyde Metz, 1984-01-01 *Study Guide to Accompany Calculus for the Management Life and Social Sciences* *Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl* Paul B. Kelter, 1993 **Study Guide [to] Chemistry, a Systematic Approach [by] Sisler, Dresdner, and Mooney** William H. Myers, William T. Mooney, 1980 **Study Guide for Introductory Chemistry** Iris Stovall, 1990 *Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e* Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological

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