

## प्रश्न

- आपको तीन परखनलियाँ दी गई हैं। इनमें से एक में अवशिलत जल एवं सोप दो में से एक में अम्लीय विलयन तथा दूसरे में क्षारीय विलयन है। यदि आपको केवल लाल लिटमस पत्र दिया जाता है तो आप प्रत्येक परखनली में रखे गए पदार्थों की पहचान कैसे करेंगे?



## 2.1 अम्ल एवं क्षारक के रासायनिक गुणधर्म समझना

### 2.1.1 प्रयोगशाला में अम्ल एवं क्षारक

#### क्रियाकलाप 2.1

- विज्ञान की प्रयोगशाला से हाइड्रोक्लोरिक अम्ल (HCl), सल्फ्यूरिक अम्ल ( $H_2SO_4$ ), नाइट्रिक अम्ल ( $HNO_3$ ), ऐसीटिक अम्ल ( $CH_3COOH$ ), सोडियम हाइड्रॉक्साइड (NaOH), कैल्शियम हाइड्रॉक्साइड [ $Ca(OH)_2$ ] फॉस्फोरिक हाइड्रॉक्साइड (KOH) मैग्नीशियम हाइड्रॉक्साइड [ $Mg(OH)_2$ ] एवं अमोनियम हाइड्रॉक्साइड ( $NH_4OH$ ) के विलयनों के नमूने एकत्र कीजिए।
- ऊपर दिए गए प्रत्येक विलयन की एक बूँद साब पत्र पर जड़ी-जड़ी से रखिए एवं सारणी 2.1 के अनुसार निम्नलिखित सूचकों से उसकी जाँच कीजिए।
- लाल लिटमस, नीले लिटमस, फेनॉल्फ्थैलिन एवं मेथिल ऑरेंज विलयन के साथ लिए गए विलयन के रंग में क्या-क्या परिवर्तन होते हैं?
- अपने प्रेक्षण की सारणी 2.1 में लिखिए।

सारणी 2.1

विलयन का नमूना	लाल लिटमस विलयन	नीला लिटमस विलयन	फेनॉल्फ्थैलिन विलयन	मेथिल ऑरेंज विलयन

रंग में परिवर्तन के द्वारा यह सूचक हमें बताते हैं कि कोई पदार्थ अम्ल है या क्षारक। कुछ ऐसे पदार्थ होते हैं जिनकी गंध अम्लीय या क्षारकीय माध्यम में भिन्न हो जाती है। इन्हें गंधीय (Olfactory) सूचक कहते हैं। आइए इनमें से कुछ सूचकों की जाँच करें।

#### क्रियाकलाप 2.2

- बारीक कटी हुई प्याज तथा खरबूट कपड़े के टुकड़े को एक प्लास्टिक के बैसे में रखिए। बैसे को कस कर बंध रखिए तथा पूरी रात क्रिज में रहने रखिए। अब इस कपड़े के टुकड़े का उपयोग अम्ल एवं क्षारक की जाँच के लिए किया जा सकता है।
- इसमें से दो टुकड़े रखिए एवं उनकी गंध की जाँच कीजिए।
- इन्हें खरबूट साठ पर रखकर उनमें से एक टुकड़े पर तनु HCl विलयन की कुछ बूँदें एवं दूसरे पर तनु NaOH विलयन की कुछ बूँदें डालिए।
- दोनों टुकड़ों को जल से धोकर उनकी गंध की पुनः जाँच कीजिए।

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