# Types of Chemical Reactions

Balance each of the following reactions and identify each type of reaction:

2. \_\_\_\_ 
$$NH_3+$$
 \_\_\_\_  $H_2(SO_4) \rightarrow$  \_\_\_\_  $(NH_4)_2(SO_4)$ 

3. \_\_\_\_ 
$$C_0H_0O + ___ O_2 \rightarrow ___ CO_2 + ___ H_2O$$

4. \_\_\_\_ Pb + \_\_\_ 
$$H_3(PO_4) \rightarrow$$
 \_\_\_\_  $H_2$  + \_\_\_\_  $Pb_3(PO_4)_2$ 

5. \_\_\_\_ Li<sub>2</sub>N + \_\_\_ NH<sub>4</sub> (NO<sub>3</sub>) 
$$\rightarrow$$
 \_\_\_ LiNO<sub>3</sub> + \_\_\_ (NH<sub>4</sub>)<sub>3</sub>N

Identify each type of reaction:

# **Classification Of Chemical Reactions Answer Key**

**L Darling-Hammond** 

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