

Skills Worksheet

2.2 Concept Review

Section: Properties of Matter

1. **Classify** each of the following as a physical or chemical property of sulfur.

_____ a. Its density is 2.97 g/cm^3 .
_____ b. It reacts with hydrogen to form a gas.
_____ c. It is a yellow solid.
_____ d. Its melting point is 112°C .
_____ e. It combines with oxygen.

2. **Classify** each of the following as a physical or chemical property of phosphorus.

_____ a. It is a white, waxy solid.
_____ b. It burns in air.
_____ c. Its melting point is 44.1°C .
_____ d. It has a density of 1.82 g/cm^3 .
_____ e. Its boiling point is 280.3°C .

3. **Explain** why aluminum is a suitable material to use in making cans based on its physical and chemical properties.

4. **Calculate** the mass of a sample of pure silver (density = 10.49 g/cm^3) that has a volume of 12.99 cm^3 .

5. **Compute** the density of an 820 g sample of pure silicon occupying a 350 cm^3 container.

Concept Review Section Physical Properties Of Solution

Yi-Tong Ma



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EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS *Solvents and Solutions: Structure and Properties* Keshra Sangwal, 2021-08-06 A UNIQUE BOOK ON THE PRESENT STATUS OF SOLVENTS AND SOLUTIONS WITH IMPORTANT PROBLEMS RELATED TO THEIR STRUCTURE AND PROPERTIES The literature on the properties of solvents and solutions used in academic research and in a wide range of industries has grown enormously during the last four decades and is scattered in different specialized journals Solvents and Solutions is a groundbreaking text that offers a systematic compilation of important problems related to selected properties of solvents and solutions based on the literature published so far The author places emphasis on explaining the basic concepts involved in understanding the properties and behavior of various solvents and solutions of electrolytes and nonelectrolytes in a consistent manner After a description of the general characteristics of structure of solvents and solutions and the solubility of electrolytes and nonelectrolytes under normal temperature and pressure conditions the book first deals with different aspects of the density and the refractive index of solvents and dilute as well as concentrated solutions and finally with the transport i e viscosity and electric conductivity and thermal properties of solvents and solutions Solvents and solutions is the first text devoted to the description and discussion of their properties since the publication of a monograph on the physical properties of aqueous electrolyte solutions more than three decades ago The main features of this book are Reflects developments in the investigation of solvents and solutions during the last three decades Outlines basic concepts involved in understanding the properties and behavior of solvents and solutions Describes and discusses different properties of ionic liquids as solvents and the behavior of their mixtures with other commonly used solvents Contents of different chapters are not only self contained but the contents are practically independent of each other Written as a practical guide for researchers who are looking for an uptodate overview of the physical and transport properties of solvents and solutions and as a reference source for workers in chemical industries and related fields and for graduate students of chemical engineering and physical chemistry **The Physical properties of colloidal solutions** Eli Franklin Burton, 1921 *Holt Chemistry* Holt Rinehart & Winston, 2003-01-24 **CliffsNotes CSET: Multiple Subjects with CD-ROM, 3rd Edition** Stephen Fisher, Jerry Bobrow, 2012-04-03 A new edition of one of the bestselling CSET products on the market Reflects the latest changes in the California CSET Multiple Subjects teacher certification test which is now computer based only The book includes diagnostic tests for every domain included in the test detailed subject review chapters and 2 full length practice tests with in depth answer explanations The CD contains all of the book s subject review chapters in searchable PDF format the book s 2 practice tests plus a third full length practice test **Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed** , 2000 Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and

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