

Class assessment:

2<sup>nd</sup> April, 2023  
Sunday

Syllabus  $\rightarrow$  Reversible reactions & Redox reactions  
+ Haber process + Contact process + fertilizers

## Reversible Reactions

These reactions can go either side.  
 $\rightarrow$  Reversible reactions are shown by the symbol " $\rightleftharpoons$ "

Equilibrium OR dynamic equilibrium:

A state in a reversible chemical reaction where concentration of reactants and products become constant (No longer changing)

A reversible chemical reaction is said to be at equilibrium state when the rate of forward reaction becomes equal to rate of backward reaction.

"Equilibrium is only attained in closed system"

# Concept Review Section Reversible Reactions Answers

**Sebastian Brünink**



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