

# Concept-Development Practice Page

6-3

## Racing Day with $a = F/m$

In each situation below, Cart A has a mass of 1 kg. Circle the correct answers (A, B, or Same for both).

1. Cart A is pulled with a force of 1 N.  
Cart B also has a mass of 1 kg and is pulled with a force of 2 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



3. Cart A is pulled with a force of 1 N.  
Cart B has a mass of 2 kg and is pulled with a force of 2 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



5. This time Cart A is pulled with a force of 4 N.  
Cart B has a mass of 4 kg and is pulled with a force of 4 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



2. Cart A is pulled with a force of 1 N.  
Cart B has a mass of 2 kg and is also pulled with a force of 1 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



4. Cart A is pulled with a force of 1 N.  
Cart B has a mass of 3 kg and is pulled with a force of 3 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



6. Cart A is pulled with a force of 2 N.  
Cart B has a mass of 4 kg and is pulled with a force of 3 N.  
Which undergoes the greater acceleration?

(A) (B) (Same for both)



thank to Dean Baird

# Concept Development Practice Page Conservation Of Energy

**Food and Nutrition Information Center  
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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Concept Development Practice Page Conservation Of Energy**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web in this set of practice questions we will summarize the main concepts of thermochemistry such as the relationship between internal energy work and heat exothermic and endothermic process heat capacity constant pressure calorimetry constant volume calorimetry the enthalpy the standard enthalpies of formation and their use in

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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web we can express the first law in many ways one of the more useful expressions is that the change in internal energy  $e$  of a system in any process is equal to the heat  $q$  added to the system plus the work  $w$  done on the system by its surroundings  $e$

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