



Delhi Public School, Prayagraj

HALF YEARLY EXAMINATION (2022-2023)

SUBJECT: - CHEMISTRY

CLASS: - XI

Duration: 3 Hours

Maximum Marks: 70

Name: - ~~XXXXXXXXXX~~

Class/Sec.: - ~~XI/A~~

Instructions:

Q.No. 1 to 10 carries 1 mark each.

Q.No. 11 to 19 carries 2 marks each

Q.No. 20 to 28 carries 3 marks each

Q.No. 29 to 31 carries 5 marks each.

- Which of the following compounds has hydrogen with oxidation number +1?
(a) HBr (b) NaOH (c) LiH (d) H₂
- In the compounds KMnO_4 and $\text{K}_2\text{Cr}_2\text{O}_7$, the highest oxidation state is of the element:
(a) Mn (b) K (c) Cr (d) O
- If travelling at same speeds, which of the following matter waves have shortest wavelength?
(a) Proton (b) Neutron (c) Alpha particle (d) Electron
- Which of the following pairs of gases contains the same number of molecules?
(a) 16 gm of O_2 and 14 gm of N_2 (b) 8 gm of O_2 and 22 gm of CO_2
(c) 28 gm of N_2 and 22 gm of CO_2 (d) 32 gm of O_2 and 32 gm of N_2
- The number of water molecules present in a drop of water (volume 0.0018 ml) at room temperature is:
(a) 6.023×10^{23} (b) 1.084×10^{24} (c) 4.84×10^{23} (d) 6.023×10^{23}
- Which of the following is true?
(a) According to Pauli's exclusion principle, no two electrons in an atom can have the same values of quantum numbers n, l and m.
(b) The total energy of an electron in an orbital is half of its potential energy.
(c) The speed of an electron in an orbit increases with increase of quantum number n.

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Table of Contents Dps Exam Sample Papers

1. Understanding the eBook Dps Exam Sample Papers
 - The Rise of Digital Reading Dps Exam Sample Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Dps Exam Sample Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Dps Exam Sample Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Dps Exam Sample Papers
 - Personalized Recommendations
 - Dps Exam Sample Papers User Reviews and Ratings
 - Dps Exam Sample Papers and Bestseller Lists
5. Accessing Dps Exam Sample Papers Free and Paid eBooks
 - Dps Exam Sample Papers Public Domain eBooks
 - Dps Exam Sample Papers eBook Subscription Services
 - Dps Exam Sample Papers Budget-Friendly Options
6. Navigating Dps Exam Sample Papers eBook Formats

- ePub, PDF, MOBI, and More
- Dps Exam Sample Papers Compatibility with Devices
- Dps Exam Sample Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Dps Exam Sample Papers
 - Highlighting and Note-Taking Dps Exam Sample Papers
 - Interactive Elements Dps Exam Sample Papers
- 8. Staying Engaged with Dps Exam Sample Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Dps Exam Sample Papers
- 9. Balancing eBooks and Physical Books Dps Exam Sample Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Dps Exam Sample Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Dps Exam Sample Papers
 - Setting Reading Goals Dps Exam Sample Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Dps Exam Sample Papers
 - Fact-Checking eBook Content of Dps Exam Sample Papers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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